

Digital Light Processing (DLP) of Nano Biphasic Calcium Phosphate Bioceramic for Making Bone Tissue Engineering Scaffolds

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Abstract

Nano-sized ceramic particles are beneficial for the properties of 3D printed structures via digital light processing (DLP). However, achieving optimal conditions for the preparation and processing of ceramic nanoparticle-based slurry is a significant challenge. In this work, a stable and well dispersed slurry with nano-sized biphasic calcium phosphate (BCP) powders was developed and porous BCP bioceramic scaffolds with good bioactivity were accurately fabricated via the DLP additive manufacturing technology. Effects of dispersant, photoinitiator and solid loading on rheological properties and curing abilities of BCP slurries were systematically studied. The slurry having 65 wt. % of nano-sized BCP powders showed a relatively low viscosity of 400 mPa·s at the 50 s⁻¹ shear rate. It was found that the optimal photoinitiator concentration was tightly associated with the energy dosage in DLP. Furthermore, the effects of sintering temperature (1100°C, 1200°C and 1300°C) on the surface morphology, shrinkage, phase constitution, hardness, compressive properties, and *in vitro* biological performance of DLP-formed bioceramics were comprehensively investigated. The results showed that DLP-formed BCP ceramics sintered at 1300°C had the best mechanical properties, *in vitro* cytocompatibility, and homogeneous bone-like apatite formation ability. Finally, DLP-formed 3D scaffolds with different pore sizes ranging from 300 µm to 1 mm were successfully fabricated, which exhibited high fidelity and accuracy. The current study has demonstrated the great potential of using DLP technology to construct functional complex BCP bioceramic scaffolds for bone tissue regeneration.

Keywords: nano-sized, biphasic calcium phosphate, digital light processing, sintering temperature

1. Introduction

Bone repair and regeneration has been in imperative demand owing to the increase of bone injuries caused by trauma, bone loss, and even congenital malformation, particularly for the defects that greater than the critical-sized bone defect (CSBD) which is difficult to heal spontaneously[1]. Bone tissue engineering (BTE) has flourished in recent years and showed great perspectives in biomedical field [2, 3]. Scaffold, as an essential component of scaffold-based tissue engineering, can give temporary structural support for cell attachment, growth, and proliferation, until the body has restored the basic functions of the host tissues[4]. So far, different types of materials have been tried for scaffolds, from metals, ceramics, and polymers to various composites[5, 6]. Among them, biphasic calcium phosphate (BCP), a mixture of hydroxyapatite (HAp) and β -tricalcium phosphate (β -TCP), is an excellent bioceramic for hard tissue repair due to its hybrid advantages of HAp and β -TCP[7, 8]. HAp is chemically similar to the inorganic composition of native hard tissues such as bone and teeth. It processes excellent biocompatibility, bioactivity, and osteoconductivity, thus is widely used in clinical bone repair [9-12]. However, the biodegradation rate of HAp is often slow, which leads to a long degradation period and restricts the bone regeneration process[13]. β -TCP is a popular bioresorbable bioactive ceramic material that can be absorbed and transformed into new bone tissues though continuous cell metabolism[14, 15]. Therefore, the combination of HAp and β -TCP has been considered as a desirable biomaterial for bone tissue engineering scaffold and numerous studies have demonstrated their efficacy[16-18].

Liu *et al.* fabricated BCP scaffolds with controlled macro-pore size and porosity and the *in vivo* implantation exhibited that they had excellent mechanical properties and bone-forming abilities [19]. Bouler *et al.* detailedly reviewed the biological responses of BCP-based materials for biomedical use and demonstrated the great future applications as osteoconductive scaffold[7].

For bone tissue engineering scaffolds, several requirements should be fulfilled, such as biocompatibility, biodegradability and sufficient mechanical property [20]. Furthermore, scaffolds should also have highly porous structures with interconnected pores, which are beneficial for nutrient and oxygen transport, thus promoting tissue regeneration. Conventional fabrication methods such as thermally induced phase separation, freeze drying, solvent casting, and electrospinning have been investigated for manufacturing tissue engineering scaffolds [21, 22], however, scaffolds fabricated by these conventional approaches present inevitable problems, including uncontrollable pore size or incomplete pore interconnectivity [23].

Recently, additive manufacturing(AM), also popularly named 3D printing, has emerged and considered as a powerful platform for fabricating tissue engineering scaffolds [24, 25]. Several AM technologies have been employed for bioceramic fabrication in tissue engineering field, such as stereolithography (SLA), binder jetting (BJ), direct ink writing (DIW), selective laser sintering (SLS), and digital light processing (DLP) [26-28]. Among these various AM technologies, DIW plays a vital role in ceramic fabrication for biomedical application owing to its low cost, easy operation, and wide materials available[29]. Alumina, zirconia, bioglass, and calcium phosphate based bioceramics have been widely reported for implant/scaffold

fabrication[30-32]. However, the resolution and dimensional accuracy are insufficient for complex structures with small features[33]. DLP is one of the most promising approaches to construct complex ceramic parts with high resolution and accuracy, as well as desirable mechanical properties[26, 34, 35].

To fabricate ceramic scaffolds via DLP, ceramic particles are added to the resin medium for the preparation of photo-curable ceramic slurry. Polymerization reactions within resin matrix will be initiated when exposed to laser or UV light. The ceramic particles will be surrounded by the cross-linked polymer in the printing[36, 37]. After thermal debinding and sintering, the polymer binders will be removed, and then pure ceramic parts will be obtained. Two key points should be considered in this process. One is to avoid the cracks caused by the thermal treatment[38]. Therefore, slurry with high solid loading is needed[39]. However, with the increase of the solid loading of ceramic slurries, the viscosity rises rapidly. It is unfavorable for the self-leveling of new layers in the printing process, which results in the printing failure, especially for the DLP machines without recoating blades[40]. Furthermore, nano-sized ceramic particles are proved to be beneficial for improving mechanical properties of the final parts, although obtaining ceramic slurry with high solid loading has more difficulties since nano-sized particles have high specific surface area and surface energy that are easy to agglomerate with surrounding particles[41]. Therefore, developing slurries containing nano-sized ceramic particles with high solid loading and low viscosity are essential in ceramic DLP process[42]. Also, achieving optimal conditions for the DLP processing of ceramic nanoparticle based slurry remains a challenge owing to the light scattering caused by the ceramic particles[43]. The other is the effect of different

sintering temperature on the properties of final ceramic parts. For bioceramic fabrication, sintering temperature is tightly related to the surface morphology, porosity, volume density, and mechanical properties of the samples and further influences the biological behaviors[44]. Especially for BCP ceramics, phase transformation may occur in the sintering process[45, 46]. However, few studies have systematically studied the effects of sintering temperature on the properties of BCP bioceramics prepared by DLP AM technology.

In the present work, nano-sized BCP ceramic slurry with 65 wt.% solid loading was prepared and processed via DLP AM technology. The effects of dispersion, photoinitiator concentrations on the rheological properties and polymerization behaviors of the BCP slurry were studied. Subsequently, the effects of different sintering temperatures on surface morphology, shrinkage, phase stability, mechanical properties, and *in vitro* biological responses of DLP-formed bioceramics were investigated. Furthermore, high-performance DLP-formed BCP scaffolds were fabricated using the optimal sintering temperature. Our results have provided insights for the fabrication of high-performance BCP BTE scaffolds by the emerging AM technology.

2. Materials and Methods

2.1 Materials

The HAp powders (needle-like shape; diameter: 20 nm; length: 270 nm) and β -TCP powders (near-spherical shape; d₅₀: 500 nm) were provided by Nanjing Emperor Nano Material Co., LTD. 1,6-hexanediol diacrylate (HDDA) as the monomer was obtained from Shanghai Yinchang New Material Co., Ltd. Diphenyl

(2,4,6-trimethylbenzoyl) phosphine oxide (TPO, Aladdidin) was used for initiating the photo-polymerization. Three types of dispersants, BYK111 (phosphoric acid ester), BYK2055 (Modified polyacrylate) (BYK Chemie, Germany), and monoalcohol ethoxylate phosphate (MAEP) (Kelon Chemical Reagent Factory, Chengdu, China) were selected for surface modification of BCP powders. Fetal bovine serum (FBS), Dulbecco's Modified Eagle Medium (DMEM), and penicillin-streptomycin were purchased from Gibco (Thermo Fisher Scientific, USA). Live-dead assay kit was provided by Invitrogen (Thermo Fisher Scientific, USA). All materials and reagents were used as received without further processing.

2.2 Surface modification of BCP powders

Three types of dispersants with the concentration of 5.0 % relative to the weight of BCP powders were applied for surface modification of the ceramic particles. Same weight of HAp and β -TCP powders (HAp: β -TCP=1:1) were dried at 80°C for 12h to avoid any moisture. Then, the dispersant was added to ethanol solution. After full dissolution of the dispersant, ceramic particles were mixed thoroughly with the dispersant solution via a turbula shaker mixer (Turbula T2F, Basel, Switzerland) for 12h. The ceramic solution was then dried in an oven at 60°C for 48h. After crushing and screening with a 60-mesh sieve, the modified BCP powders were obtained. The rheological properties of 40 wt. % BCP slurry with modified powders were evaluated and compared. Afterwards, the better dispersant with different concentrations (1.0%, 3.0%, 5.0%, 8.0%, 9.0%, 10.0%, 12.0%, and 15.0%) relative to the weight of BCP powders were prepared in the same way to obtain the optimal ratio of the dispersant for BCP powder modification.

2.3 TPO concentration optimization and slurry preparation

Photoinitiator is an essential component in the photo-curable slurry. It can absorb sufficient energy to initiate the polymerization and thus have a close relationship with the curing abilities of the slurry. Therefore, the effect of TPO concentration on cure depth was investigated. Different TPO amounts ranging from 0.5 wt. % to 2.0 wt. % relative to the mass of resin weight were employed for BCP slurries with 40 wt. % solid loading.

Then, the photo-curable resin was prepared by mixing HDDA and TPO with specific concentration. Whereafter, modified BCP powders were added in the resin matrix and mixed by planetary ball milling machine with the speed of 400r/min for 12h to obtain stable and homogenous slurry. In this context, suspensions with different solid loading (50 wt.%, 60 wt%, 65 wt%, 70 wt.%, and 75 wt.%) were prepared [40].

To obtain the optimal printing parameters, curing abilities of the ceramic slurry should be evaluated. Cure depth reflects the curing behavior under the given energy and is required several times larger than the layer thickness to avoid insufficient interlayer combination. Generally, the cure depth is influenced by the total energy input according to the Jacobs equation[36]:

$$C_d = D_p \ln\left(\frac{E}{E_c}\right) - - - - - (1)$$

where C_d is the cure depth, D_p is the penetration depth when the light intensity is reduced to 37%, E is the energy input, E_c is the critical energy. Based on the experimental data of C_d at different E , the relation between C_d and E can be

obtained through linearly fitting to Jacobs equation. That equation could provide a guidance for parameter selection in the printing.

2.4 Fabrication of BCP ceramic parts via DLP

The BCP green parts were fabricated by a commercial bottom-up DLP printer (Asiga Max, Australia), which is capable of adjusting the energy density, exposure time, and layer thickness. First, the standard stereolithography (STL) file was prepared and transferred. When printing started, the printing platform positioned one layer thickness above the bottom of the resin tank. Second, the UV light selectively exposed to the resin surface and cured one-layer slurry based on the slicing data. Third, the printing platform uplifted and made the cured parts detach with the transparent substrate. The platform lowered again after the slurry self-leveling and started the next-layer printing. The process repeated layer by layer, until the entire part was fully completed. Finally, the green body was ultrasonically cleaned with ethanol to remove the uncured slurry.

To determine the optimal debinding parameters and obtain the crack-free BCP scaffolds, thermogravimetric analysis (TGA) (Discovery, TA Instruments, USA) was conducted with heating from 30°C to 700°C in a nitrogen atmosphere at a rate of 10°C/min. Based on the TGA result, an optimized debinding and sintering strategy were used to obtain the final BCP bioceramic in a muffle furnace, the scheme of BCP fabrication via DLP AM technology is shown in Fig.1.

2.5 Characterization

The rheological property was performed via a rheometer (MCR-92, Anton Paar, Australia). Shear rate ramped from 0 s⁻¹ to 100 s⁻¹. The cure depth was obtained

through controlling the first exposure and measured by a digital micrometer (Mitutoyo MD-25PX, Japan). The surface microstructure and morphology of sintered ceramics were observed by a scanning electron microscopy (SEM, ZEISS Merlin, Germany) with an acceleration voltage of 5KV. The phase composition was confirmed by X-ray diffraction (XRD, Rigaku Smartlab, Japan) at a scanning rate of 10°/min from 20° to 60°. The mass fraction of the main phase was calculated via adiabatic method. If the samples include two phases A and B, the mass fraction can be calculated by the following equation (2)[47]:

$$W_A = \frac{I_A}{I_A + \frac{I_B}{K_A^B}} ; W_B = \frac{I_B}{I_B + I_A * K_A^B} ; K_A^B = \frac{RIR_B}{RIR_A} \text{---(2)}$$

Where W is the mass fraction, I is the integrated intensity of the strongest peak of the phase (the area of the strongest peak), RIR is the reference intensity ratio and can be found in the PDF card ($RIR_{HAp} = 1.10, RIR_{\beta-TCP} = 1.20$) [48].

The shrinkage was calculated according to the equation (3)[49]:

$$S = \frac{Y - L}{L} * 100\% \text{---(3)}$$

where S is the linear shrinkage, L is the size after sintering, Y is the designed size of the model.

The open porosity of scaffolds was determined using the Archimedes method by employing an electronic analytical balance (FA2104J, Sunny Hengping Scientific Instrument, Shanghai). The weight of scaffolds in the dry state was measured as M_d . Then, the scaffolds were immersed in ethanol in a beaker which was put in a vacuum chamber for 10 min to force out the trapped air in the pores. The scaffolds were

removed from ethanol and filter paper was used to remove the ethanol from the surface. The weight of the wet scaffolds was measured in the air as M_w while M_l denoted the weight of the submerged scaffold. The porosity was calculated by the equation below [50]:

$$Porosity(\%) = \frac{M_w - M_d}{M_w - M_l} * 100\% \quad (4)$$

The bulk volume density of the sintered disk ($\Phi 10 \times 2 \text{ mm}$) was obtained following the equation (5):

$$Volume\ density = \frac{M_d * \rho_l}{M_l - M_d} \quad (5)$$

Where ρ_l is the density of the liquid (ethanol). Five samples were tested for open porosity and volume density for an average, respectively.

Compressive strength of porous cubic structure with 58 vol.% porosity (dimensions: $5.4 \times 5.4 \times 5.4 \text{ mm}$, pore size: 1 mm; square pore) was studied by a universal testing machine (TST-C1019, 0-2 kN, China) with the loading speed of 1mm/min. Five samples were tested for an average. The hardness of the solid disks ($\Phi 10 \times 2 \text{ mm}$) was evaluated through Vickers hardness tester (HXD-1000TMC/LCD, China) and the load was 200 g with dwell time 20 s. Five samples and five regions for each sample were tested. The surface roughness of the sintered samples ($\Phi 10 \times 2 \text{ mm}$) was characterized through a laser scanning confocal microscope (VK-X1000, KEYENCE, Japan). Three samples were tested in five regions.

2.6 *In vitro* biological evaluation of BCP bioceramics

Rat bone mesenchymal stem cells (BMSCs) derived from bone marrow were used to evaluate the biological behavior of DLP-formed BCP bioceramics (bulk disks:

Ø6×1 mm). To explore the cell proliferation behavior on BCP ceramics, BMSCs at a density of 1.0×10^3 cells/well were seeded on three types of BCP ceramics and blank culture plate (control group) in 96-well cell culture plate and cultured with DMEM medium in an incubator supplemented with 5% CO₂ at 37°C. After culturing for 1, 3, and 5 days, the cell viability was performed by MTT assay.

To evaluate cell survival and attachment on the BCP bulk disk samples, BMSCs at a density of 5.0×10^3 cells/well were seeded. After incubation for 4 h and 24 h, the live/dead assay was used to study the cell survival and adhesion on BCP ceramics. After staining, the cells were observed with a fluorescence microscope (Leica DMI8, Germany). The living cells were stained green, and the dead cells were red when exposed to relative fluorescence. Besides, after 24 h of culture, BCP samples were fixed by 4% polyformaldehyde (Sigma-Aldrich, USA) for 20 min and then washed with PBS twice. Subsequently, BCP samples were dehydrated in a graded ethanol series from 30, 50, 60, 70, 80, 90, 100% ethanol for 10 min each. The samples were then dried at room temperature, and then sputter coated with gold for SEM observation.

2.7 Bioactivity evaluation

To evaluate the bioactivity of the DLP-formed BCP bioceramics, the samples were soaked in the simulated body fluid (SBF, Shanghai yuanye Bio-Technology Co., Ltd, Shanghai, China) for 72 h in 37°C water bath. Afterwards, the samples were washed with deionized water and dried at 60°C for 6 h. Then, SEM was used to observe the apatite formation on the ceramic surface. Besides, the chemical composition towards the formation layer was investigated by Energy Dispersive X-

Ray Spectroscopy (EDS).

2.8 Statistical analysis

All quantitative data were obtained through three or five parallel samples and presented as mean value \pm standard deviation. The hardness, compressive strength, and MTT assay result were analyzed using one-way ANOVA. Significant difference was set as $p < 0.05(*)$.

3. Results and discussions

3.1 Study of the dispersants

To obtain a stable and well dispersed BCP ceramic slurry, the dispersant plays a key role, which was studied comprehensively in this work. Fig.2 shows the viscosities of 40 wt.% solid loading BCP ceramic slurries with (a) different dispersant types and (b) BYK111 with various concentrations ranging from 1.0 wt.% to 15.0 wt.%. Obviously, slurries with BYK111 modified powders had the lowest viscosity compared to other dispersants, indicating it should be used for the following preparation of the photo-curable slurry. This is might due to the anchoring head of the phosphate group in the BYK111, making it easily combine with the BCP particles while polymer chain at the other side provides the affinity to non-aqueous media. It leads to a thick steric barrier layer on the surface of the ceramic particles and forms the sufficient repulsive force between the surrounding ceramic powders, which exhibits the best dispersion effects [51, 52]. Furthermore, it was observed that the optimal ratio for BYK111 modified BCP ceramic powders was 9.0 wt. % from Fig.2(b). When the dispersant amount was less than 9.0 wt. %, the ceramic powders were insufficient dispersed and had a higher viscosity. However, when the dispersant

exceeded the optimal ratio, slurry viscosities slightly increased, which indicated the rheological property of the slurry deteriorated. This might be attributed to the addition of excessive dispersants. If the unreacted dispersants overcame a threshold, they remained free in the slurry and a flocculation related to osmotic effects could occur which damaged the stability of the slurry and increased the viscosity [53, 54].

3.2 The optimal amounts of TPO and properties of ceramic slurries

The effect of TPO concentration on the cure depth of the slurry at different energy input is shown in Fig.3 (a). The cure depth was positively correlated with energy dose for a given TPO concentration owing to the high energy absorbed of the system. Besides, as the energy dose increased, the location of the optimal TPO concentration moved forwards lower concentration. The results showed a good agreement with Lee's work[55]. Lee *et al.* investigated the cure depth of pure resin system as a function of photoinitiator concentration in both experimental and theoretical side. The proposed equation illustrated that the critical photoinitiator concentration was closely associated with the energy dosage. With the photoinitiator concentration raising, the penetration depth of the photons decreases, thus reducing the free radical initiation. The current results indicated that the addition of the ceramic particles did not affect this rule and can be applied to the ceramic slurry system for DLP printing.

Fig.3 (b) shows the rheological behaviors of 50 wt.%, 60 wt.%, 65 wt.%, 70 wt.%, and 75 wt.% BCP ceramic slurries as the function of shear rates. All these slurries showed a shear thinning behavior that was beneficial for the printing and

storing. In this study, 65 wt.% solid loading was employed for the following printing owing to its fast self-leveling speed and sufficient printing accuracy. The optimized nano-sized BCP slurry with 65 wt.% solid loading showed competitive lower viscosity of 400 mPa·s at 50 s⁻¹ shear rate [9, 56]. The curing ability of 65 wt.% slurries was studied as shown in Fig.3(c). After data fitting to the Jacobs equation, the relationship between cure depth and exposure energy for 65 wt.% BCP slurry can be described by equation (6), which can be used for selecting suitable printing parameters for the printing.

$$C_d = 150.1 * \ln \left(\frac{E}{5.0} \right) \quad \text{--- --- --- --- ---} \quad (6)$$

Considering the cure depth-energy dose equation, 20 mJ/cm² energy dose with TPO concentration 1 wt.% relative to the mass of resin weight was employed in this study. The final formulation of BCP slurry contained 65 wt.% BCP powders (HAp: β -TCP=1:1), 5.85 wt.% BYK111 (9 wt.% relative to weight of BCP powders), 28.86 wt.% HDDA, and 0.29 wt.% TPO (1 wt.% relative to weight of resin).

TGA was performed to analyze the thermal decomposition of the green part and guide the profile of the debinding process, as shown in Fig. 4(a). It can be found the mass loss mainly happened between 300 and 500 °C. The 435°C peak indicated the fast decomposition of the photosensitive resin. Therefore, the debinding and sintering profile was set as shown in Fig.4b. Originally, a heating rate of 1°C/min was used before 300 °C. Then, heating to 500°C at a slow heating rate of 0.15 °C/min. Afterwards, the heating rate increased to 1°C/min. All the printed samples kept at 435°C and 600 °C for 3h. After debinding, the samples of BCP ceramics were divided into three groups and sintered at 1100°C, 1200°C, and 1300°C at 2 °C/min,

respectively.

3.3 Shrinkage, density, and microstructure

The sintered ceramics fabricated via DLP process will undergo inevitable shrinkage during sintering[57]. Understanding the shrinkage and dimensional accuracy is important to predict the dimensional difference between 3D design and final products. A cylinder was tested in current work and the size of the tested samples in the X-Y and Z dimensions was shown in Table1. Shrinkages of X-Y and Z direction of the sintered-1100 BCP ceramics were 6.5% and 6.0%, respectively. After sintering at 1200°C, there was a huge increase of the shrinkage in three directions, ~16.2% in X-Y plane and 17.0% in Z direction. This indicated that the ceramic particles began to tightly interconnect with surrounding ceramic particles and the ceramics became denser when heated from 1100 to 1200°C. The SEM observation (Fig.5 a-b) and volume density value also demonstrated the shrinkage changes. After sintering at 1300°C, the sintered BCP ceramics became much denser without any voids and the particles were tightly interconnected with all grains (Fig.5c), demonstrating a high level of densification were obtained. It was shown that the shrinkage was approximately equal in three dimensions at ~22% after sintering at 1300°C, suggesting a near-isotropic shrinkage which may effectively avoid cracking in the thermal treatment. Fig.5d shows the volume density of the sintered scaffolds. The volume density of the scaffolds after sintering at 1100 °C, 1200°C, and 1300°C was 2.780 ± 0.010 , 2.935 ± 0.012 , and 3.037 ± 0.009 g/cm³, respectively. With the increase of the sintering temperature, the volume density increased obviously. This result was in good agreement with the microstructure analysis.

3.4 Surface roughness

Surface roughness of substrate is tightly related to the cell behaviors and has a direct influence on cell adhesion, proliferation, and phenotype expression [58-60]. Fig.6 shows the 3D topography images and surface roughness values of the sintered parts. The top surface is the side closed to the building platform while the bottom surface is the final cured layer. The top surface is rougher than the bottom surface because the roughness of top side depends on the quality of the building platform and a rough platform surface is beneficial for the attachment of the first layer. The bottom surface reflects the true curing surface quality of the sintered parts. Generally, the surface roughness of the sintered disks treated by three sintering temperatures (1100°C, 1200°C, and 1300°C) had no significant difference. With the sintering temperature increasing, the surface roughness of obtained parts slightly decreased. The samples sintered at 1300°C exhibited a smoother surface roughness, which was mainly owing to its denser microstructures without any voids. Hatano *et al.* suggested osteoblasts attached and proliferated more effectively when the surface roughness of the substrate was about 0.8 μm [61]. Therefore, in current study, the BCP ceramics sintered at the range of 1100°C-1300°C could be also beneficial for cell behaviors.

3.5 Phase constitution

The phase constitution of the sintered parts is closely related to the physicochemical properties of the scaffolds. Fig.7 displays the XRD pattern of the sintered BCP samples at 1100°C, 1200°C, and 1300°C. After sintering, the strong peaks of HAp (0,0,2) (2,1,1) (1,1,2) and β -TCP (2,1,4) (0,2,10)(2,2,0) were detected

without any secondary phases which were in accordance with Lee's work[62]. There were no standard peaks of α -TCP, showing the transformation of β -TCP to α -TCP was not occur[63]. The absence of the resin and dispersant demonstrated the purity of the sintered ceramic. Furthermore, the mass fraction of the HAp after sintering at 1100, 1200, and 1300°C was $54.1 \pm 0.3\%$, $46.9 \pm 1.0\%$, and $45.9 \pm 1.9\%$, respectively. As the sintering temperature increased, the ratio of HAp to β -TCP decreased, indicating the phase transition from HAp to β -TCP was occur after thermal treatment. Overall, the XRD results indicated that the HAp was not much decomposed after sintering and could offer excellent bioactivity for bone tissue engineering applications.

3.6 Mechanical properties

An ideal scaffold should have desirable mechanical properties and maintain structural integrity until the targeted tissues restore their inherent functions and properties. The hardness and compression strength are two essential factors to evaluate whether the implants or scaffolds are suitable for the loading-bear application in clinic. Fig.8(a) shows the Vickers hardness of the sintered BCP ceramic disks. It was seen that the hardness of 1300°C-sintered samples was 6.36 ± 0.43 GPa. To evaluate the compressive strength of porous BCP scaffold for BTE application, 58 vol.% porosity of the cubic-shaped scaffold was designed and printed via DLP AM technology. After printing, the porosity of the green bodies was $55.1 \pm 2.2\%$. The decrease of the porosity after printing as compared to 3D models was because of the light scattering in the photopolymerization. After sintering at three temperatures,

their porosities were $68.4 \pm 2.1\%$, $60.9 \pm 1.4\%$, and $56.3 \pm 2.8\%$, respectively. The differences were mainly attributed to the microstructure of the sintered ceramics. Fig.8(b) shows the compressive strength of the porous cubic scaffolds. It was seen that the compressive strength values of 1300°C-sintered samples reached 20.07 ± 2.09 MPa. Generally, as the sintering temperature increased, both the hardness and compressive strength dramatically increased, which indicated after sintering at 1300°C, the BCP ceramics had the best mechanical properties. This could be explained by the high densification of the BCP particles in the 1300°C-sintered samples. The compressive strength of human cancellous bone is about 1.5-45 MPa [64], and the hardness of human bone is associated with its composition, porosity, and structures. In current study, the BCP scaffolds sintered at 1300°C with about 56.3 vol.% porosity presented comparable mechanical properties with human cancellous bone, which seemed that it could be possible for bone tissue engineering application.

3.7 *In vitro* biological properties of BCP bioceramics

The BMSCs were used to evaluate the *in vitro* biological behaviors of the sintered BCP bioceramics. Fig.9 shows the cell proliferation rate of BMSCs cultured on the DLP-formed BCP ceramics for 1, 3 and 5 days. BCP samples sintered at 1100, 1200, and 1300°C showed similar cell proliferation rates with the blank control, which indicated that BCP ceramics were biocompatible and could provide a suitable environment for cell growth. Furthermore, BCP samples sintered at 1300°C showed the highest OD value in comparison with BCP samples sintered at 1100°C and 1200°C,

suggesting that BCP ceramics sintered at 1300°C performed a better cell response. This could be attributed to the higher stiffness of BCP ceramics sintered at 1300°C. Previous studies have revealed that cells were sensitive to substrate stiffness and surface morphology^{[51][65]}.

The cell survival and attachment on the sintered bulk samples after 4 h and 24 h incubation was examined by *in vitro* live/dead assay. As shown in Fig.10A, almost no red fluorescence was shown in the images, indicating that BCP ceramics were significantly biocompatible. After 4 h of incubation, compared to samples sintered at 1100°C, BMSCs presented a better spread cell morphology in the BCP samples sintered at 1200°C and 1300°C, which indicated that BMSCs could readily adhere on the surface of BCP samples sintered at 1200°C and 1300°C. After 24 h incubation, BMSCs performed well-spread cell morphology on all samples' surfaces as culture time increased. Besides, as shown in Fig.10B, the SEM observation results implied that BMSCs could maintain cell phenotype and their morphology was well expanded. BMSCs were in a flat and long shape and exhibited stretched morphologies with obvious pseudopodia [Fig.10B(b-f)], demonstrating the sintered BCP ceramics were favorable for cell attachment and adhesion.

3.8 Apatite formation ability

The formation of apatite crystals can predict the *in vivo* bioactivity of the tissue-engineering scaffold [66, 67]. This apatite layer is similar to bone mineral and can easily form chemical bonds with surrounding native tissues. Fig.11 (a,c,e) shows the surface morphology of the sintered BCP ceramics after soaking in the SBF for 3 days.

Obviously, plenty of bone-like apatite was formed on the surface of the sintered ceramics. The SBF solution was oversaturated due to the release of Ca^{2+} , HPO_4^{2-} and PO_4^{3-} ions through dissolution of BCP ceramics and formed nucleation sites on the ceramic surface. The agglomerated apatite crystals were inhomogeneously distributed on the 1100°C-sintered and 1200°C-sintered ceramics (Fig.11 a,c) while the 1300°C-sintered ceramics was completely covered with a uniform layer of apatite crystals on the surface (Fig.11e). This could be explained by the difference of surface morphology of three sintered samples[68]. Fig.11 (b,d,f) shows the chemical composition of these bone-like apatite crystals. Large amounts of Ca, P and O were detected, indicating the calcium phosphate crystals were formed. In addition, small amounts of Mg, Na and Cl were appeared mainly owing to the deposition of the SBF solution[69]. Overall, the apatite formation ability demonstrated the good bioactivities of the sintered bioceramics. The uniform apatite layer formation of the 1300°C-sintered ceramics might have a better interface with native tissues after implantation.

3.9 Fabrication of tissue engineering scaffolds

To verify the fidelity and fabrication freedom of the BCP optimized slurry and printing performance of DLP process, different designs, e.g., porous cubic structures, triply periodic minimal surface (TPMS)-G, 3D scaffolds with triangular and circular pores were designed and fabricated through the DLP printer. The pore sizes ranging from 300 μm to 1mm were successfully obtained, which exhibited the great abilities of processing complex structure with high resolution and accuracy, as shown in

Fig.12. Generally, pore size greater than 300 μm is recommended for bone scaffold owing to better vascularization and enhanced new bone formation[70]. Therefore, the fabrication ability in this study could fulfill the suitable pore size and porosity of tissue engineering scaffolds. Besides, it exhibited great competitiveness of design freedom as compared to the published works since various geometries and morphologies could be printed directly. The current study demonstrated the great potential of DLP in constructing complex BCP scaffolds for bone tissue engineering.

4. Conclusions

The nano BCP slurry with high solid loading and low viscosity was prepared and BCP bioceramics were successfully fabricated using DLP AM technology. The powder modification, slurry formulation, and DLP process optimization were systematically investigated. Furthermore, the effects of sintering temperature on the mechanical properties and *in vitro* biological properties of DLP-formed bioceramics were investigated. The conclusions include the following:

- 1) The optimal dispersant is 9% BYK111 for the surface modification of nano-sized BCP powders (at the HAp: β -TCP ratio of 1:1) to produce photo-curable slurry with high solid loading (65wt.%) as well as low viscosity(400 $\text{mPa}\cdot\text{s}$ at 50 s^{-1} shear rate level).
- 2) The optimal photo-initiator concentration is dependent on the energy dose. As the energy dose increases, the location of the optimal photo-initiator concentration

moves forwards lower concentration side.

3) The BCP bioceramic sintered at 1300°C shows improved mechanical properties as compared to those sintered at 1100°C or 1200°C. The volume density of the 1300°C-sintered BCP ceramics can reach to $3.037 \pm 0.009 \text{ g/cm}^3$ while the hardness is $6.36 \pm 0.43 \text{ GPa}$. The compressive strength of cubic-shaped porous scaffolds with ~56.3 vol.% porosity is about $20.07 \pm 2.09 \text{ MPa}$.

4) The BCP ceramics sintered at three respective temperatures show good biocompatibility. Particularly, samples sintered at 1300 °C exhibit higher cell proliferation rates, better cell spreading morphology and homogeneous bone-like apatite formation ability.

5) This work has investigated the DLP fabrication of BCP bioceramic and its scaffolds in terms of optimization of slurry formula and sintering process, which could provide a good guidance for preparing high-performance BCP bone scaffolds with high accuracy and good bioactivity using the DLP AM technology. Potentially, the methods and procedures developed in this study may be applied to the DLP fabrication of other ceramics.

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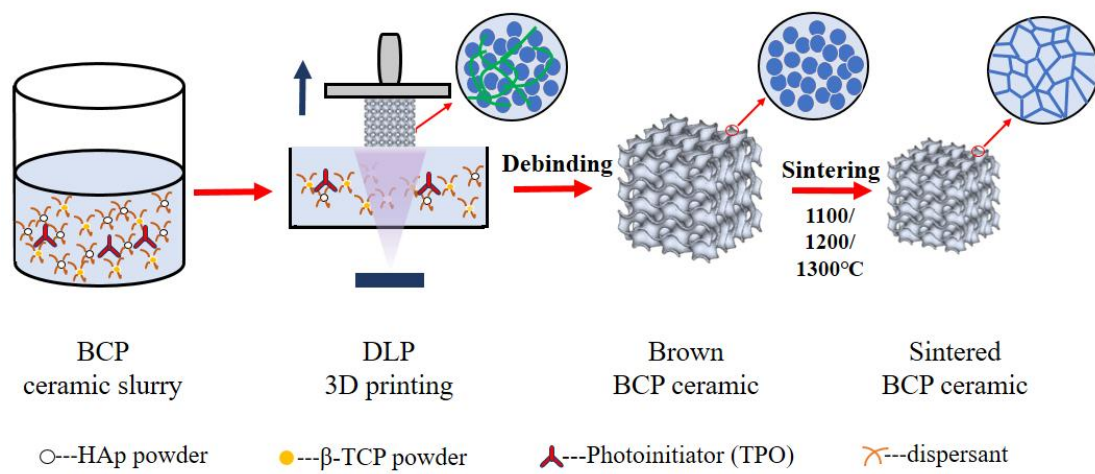


Fig.1. Schematic illustration for BCP ceramic scaffold fabrication via DLP.

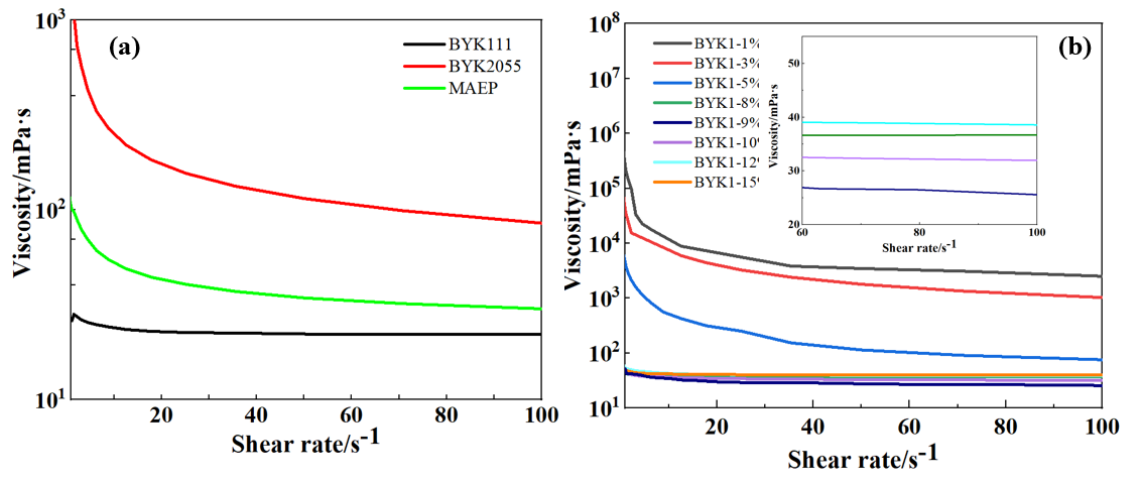


Fig.2 Effects of (a) three different types of dispersants and (b) BYK111 with different concentrations on viscosities of 40 wt. % BCP ceramic slurry.

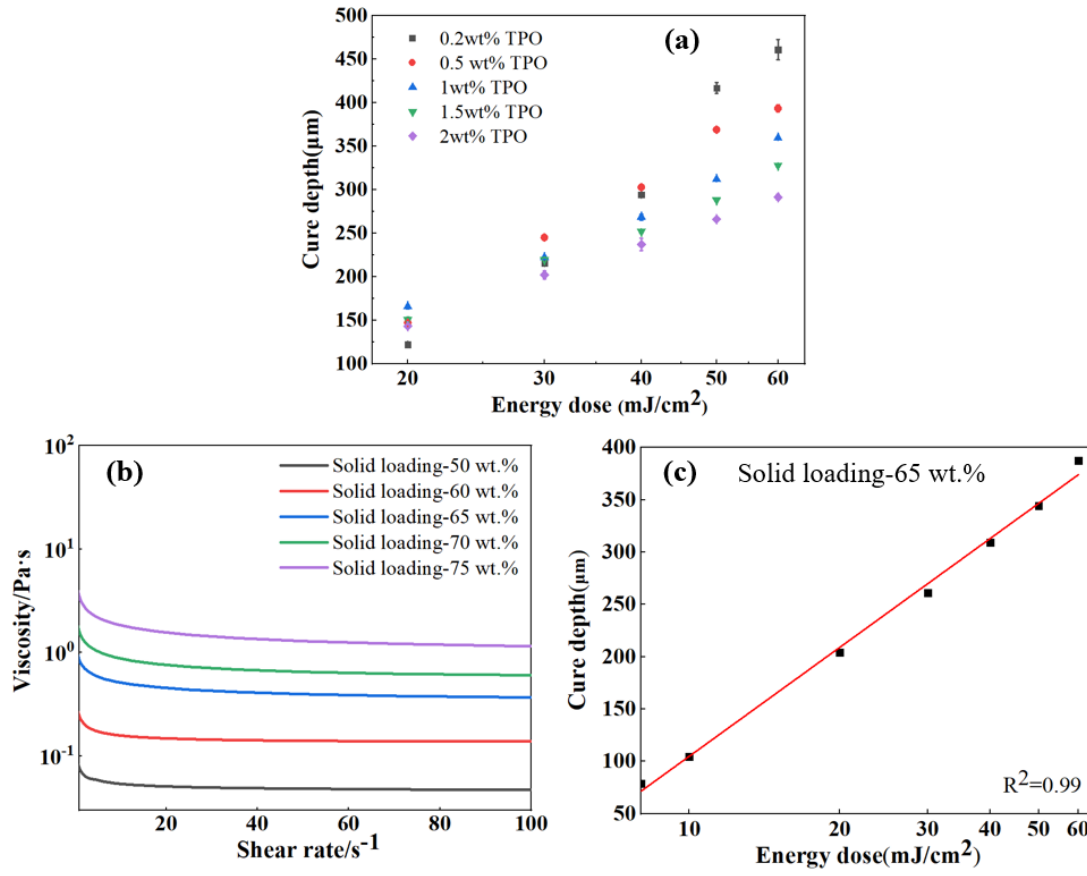


Fig.3. (a) Effects of TPO concentration and energy dose on cure depth; (b) Viscosities of BCP slurries with different solid loading; (c) The relation of cure depth and energy dose of slurry with 65 wt.% BCP.

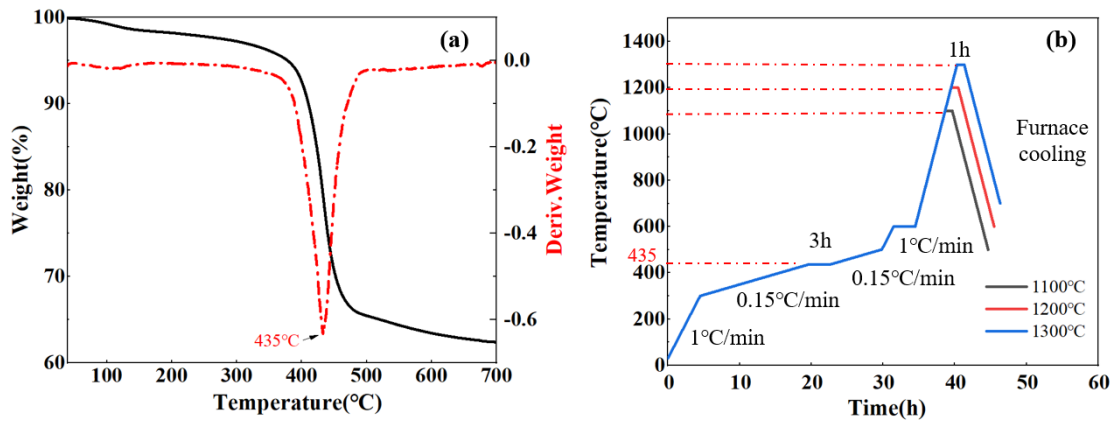


Fig.4. (a) TGA/DTG curve of green part; (b) debinding and sintering profile.

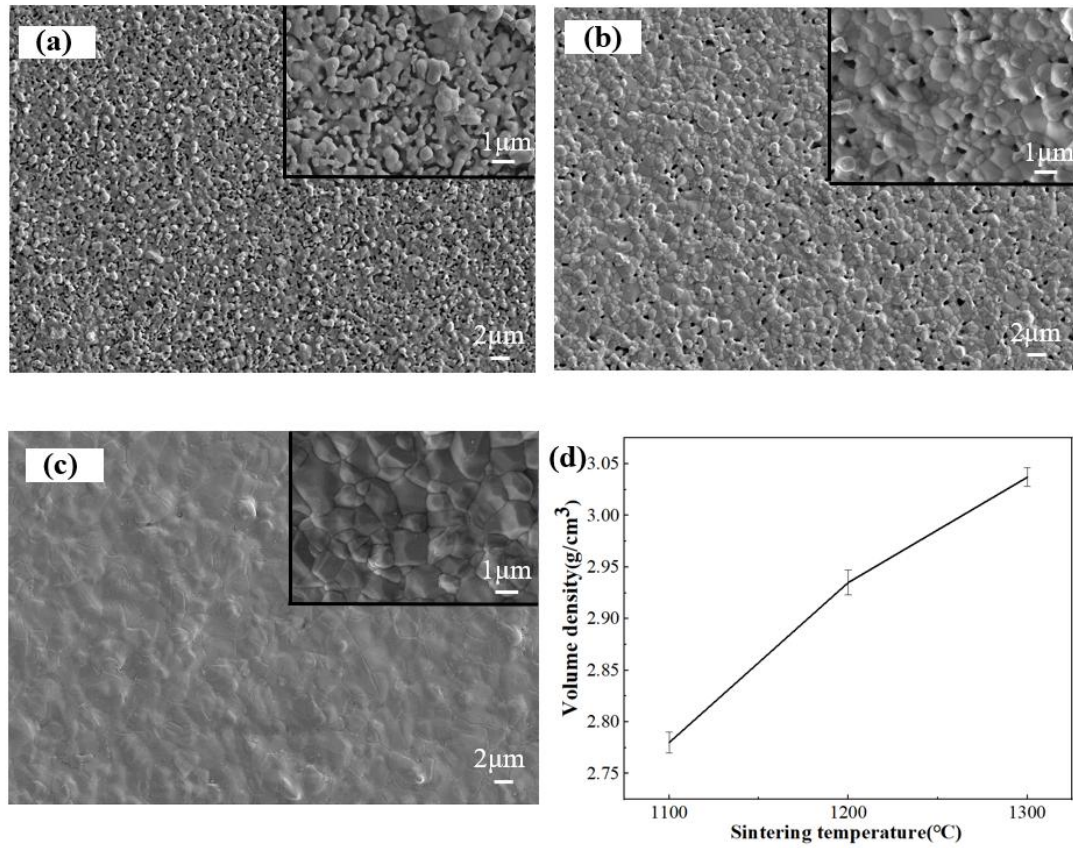


Fig.5. Microstructures of samples after sintering at (a) 1100°C; (b) 1200°C; (c) 1300°C. (d) Volume density of samples after sintering 1100-1300°C.

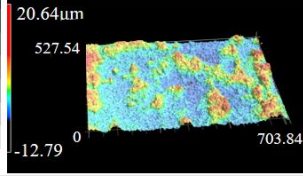
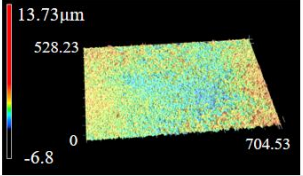
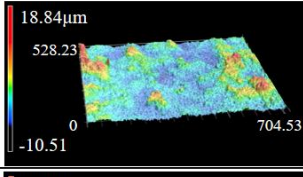
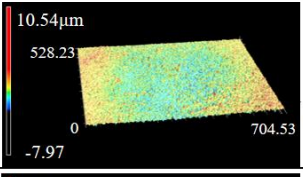
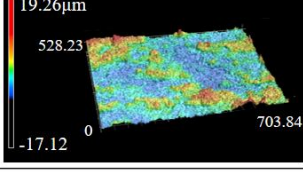
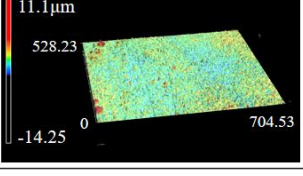
	Top Surface		Bottom Surface	
	3D images	Sa(μm)	3D images	Sa(μm)
Sintering -1100		2.53 ± 0.16		0.83 ± 0.08
Sintering -1200		2.31 ± 0.16		0.77 ± 0.10
Sintering -1300		2.31 ± 0.16		0.75 ± 0.01

Fig.6. 3D topography and surface roughness of the sintered BCP parts

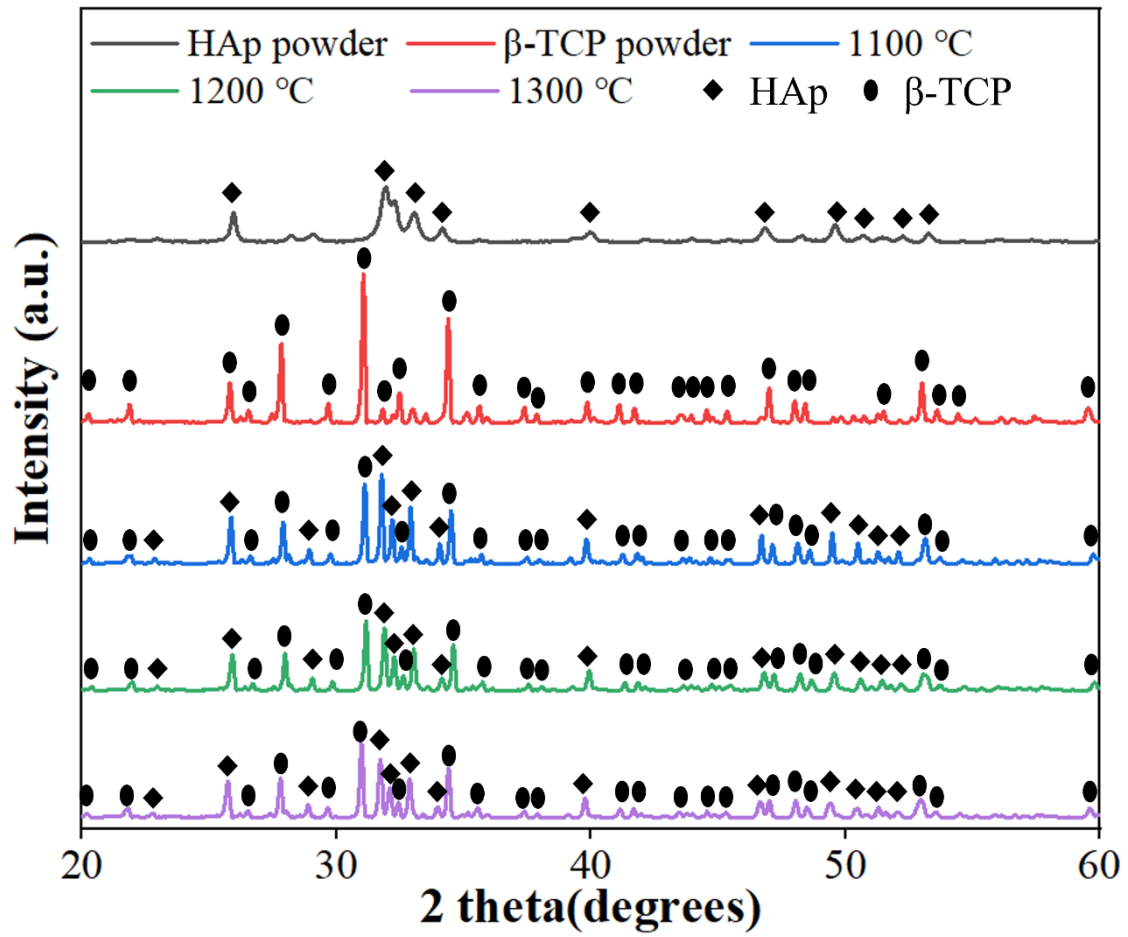


Fig.7. XRD pattern of sintered BCP ceramics after sintering at 1100°C, 1200°C, and 1300°C.

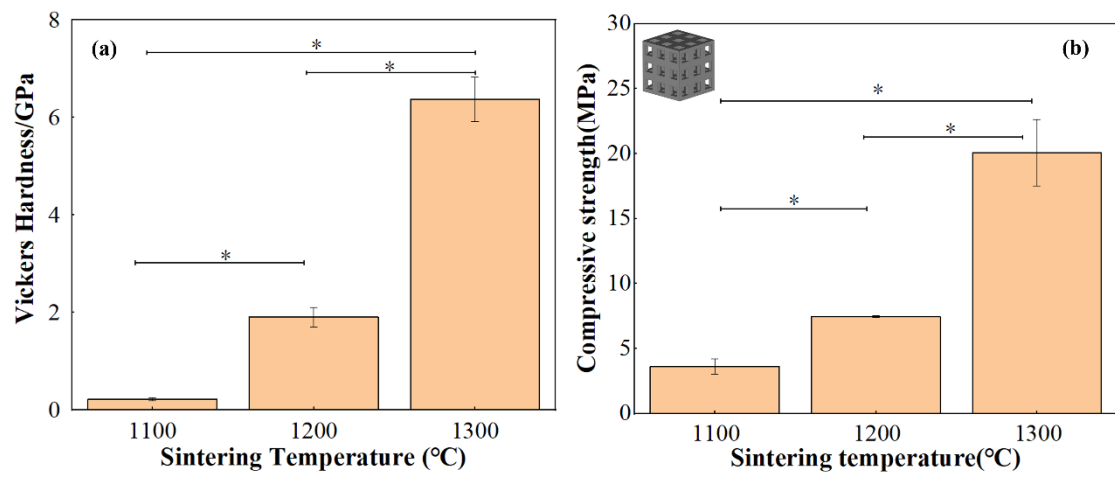


Fig.8. Mechanical properties of sintered BCP ceramics: (a) Vickers hardness; (b) Compressive strength of scaffolds (* represents $P < 0.05$)

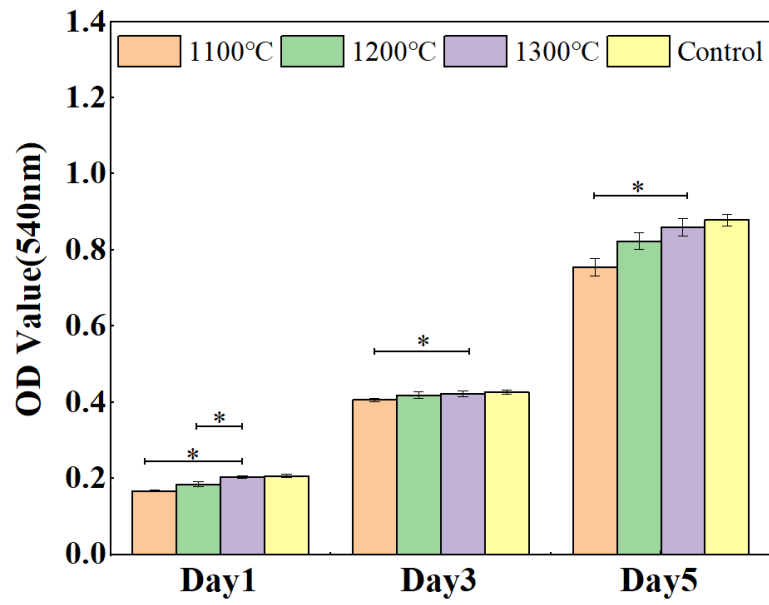


Fig.9. Results of MTT assay of BMSCs cultured on the DLP-formed BCP ceramics
(* represents $P < 0.05$)

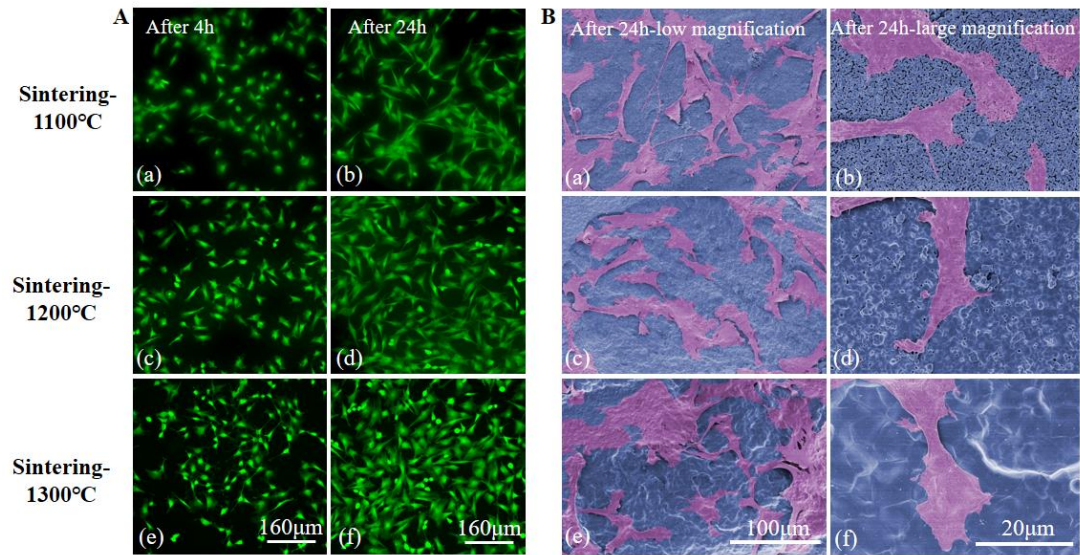


Fig.10 *In vitro* biological responses on three groups of sintered BCP ceramics: A. fluorescence images of BMSCs live/dead staining after cultured for 4h and 24h; B. SEM images of BMSCs morphology after cultured for 24h.

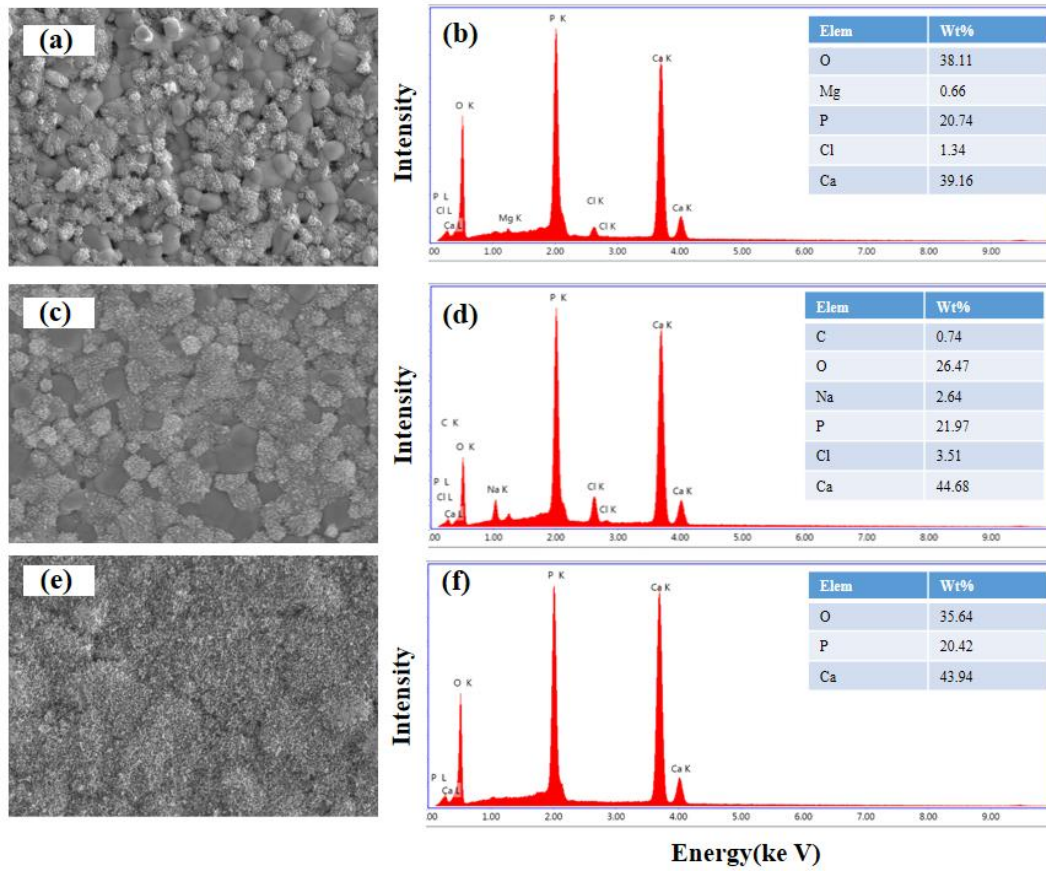


Fig.11 SEM images and EDS analysis of bone-like apatite formed on sintered BCP ceramics after soaking in SBF for 3 days: (a-b) 1100°C; (c-d) 1200°C; (e-f) 1300°C

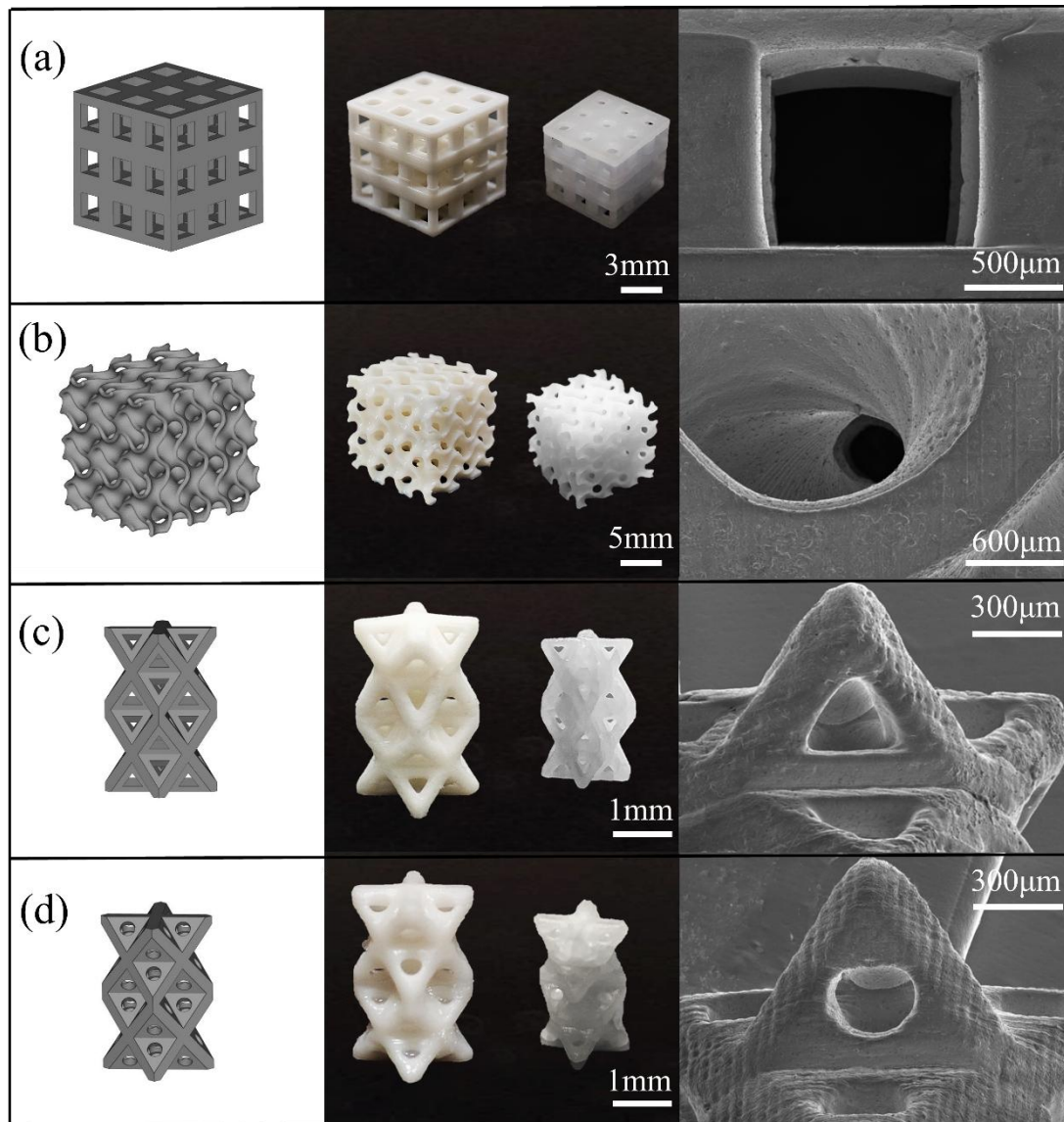


Fig.12. 3D models (left column), green bodies, 1300°C-sintered BCP scaffolds (middle column), and SEM images of various scaffolds (right column): (a) cubic (used for compression test); (b) TPMS-G; (c-d) 3D scaffolds with triangular and circular pores.

Table 1. Shrinkages of DLP-formed BCP ceramics

Material	X-Y direction		Z direction	
	Diameter/mm	Shrinkage/%	Thickness/mm	Shrinkage/%
Design	10		1	
Green body	10.37±0.05		1.08±0.03	
1100°C sintering	9.35±0.08	6.5	0.94±0.01	6.0
1200°C sintering	8.38±0.02	16.2	0.83±0.01	17.0
1300°C sintering	7.76±0.03	22.4	0.78±0.01	22.0