

1 **Intrinsic Nanoscale Structure of Thin Film Composite Polyamide**  
2 **Membranes: Connectivity, Defects, and Structure-Property**  
3 **Correlation**

4

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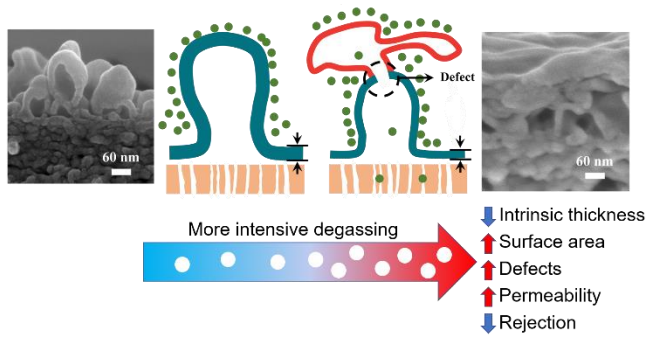
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22

23 **ABSTRACT**

24 Transport of water, solutes, and contaminants through a thin film composite (TFC)  
25 membrane is governed by the intrinsic structure of its polyamide separation layer. In  
26 this work, we systematically characterized the nanoscale polyamide structure of four  
27 commercial TFC membranes to reveal the underlying structure-property relationship.  
28 For all the membranes, their polyamide layers have an intrinsic thickness in the range  
29 of 10 - 20 nm, which is an order of magnitude smaller than the more frequently reported  
30 apparent thickness of the roughness protuberances due to the ubiquitous presence of  
31 nanovoids within the rejection layers. Tracer filtration tests confirmed that these  
32 nanovoids are well connected to the pores in the substrates via the honeycomb-like  
33 opening of the backside of the polyamide layers such that the actual separation takes  
34 place at the frontside of the polyamide layer. Compared to SW30HR and BW30, loose  
35 membranes XLE and NF90 have thinner intrinsic thickness and greater effective  
36 filtration area (e.g., by the creation of secondary roughness features) for their polyamide  
37 layers, which correlates well to their significantly higher water permeability and lower  
38 salt rejection. With the aid of scanning electron microscopy, transmission electron  
39 microscopy, and tracer tests, the current study reveals the presence of nanosized defects  
40 in a polyamide film, which is possibly promoted by excessive interfacial degassing.  
41 The presence of such defects not only impairs the salt rejection but also has major  
42 implications for the removal of pathogens and micropollutants.

43

## 44 INTRODUCTION

45 Reverse osmosis (RO) is the method of choice for seawater desalination and potable  
46 water reuse.<sup>1-3</sup> Commercially available RO membranes are prepared from an interfacial  
47 polymerization (IP) reaction between *m*-phenylene diamine (MPD) and trimesoyl  
48 chloride (TMC). These membranes have a thin film composite (TFC) structure in which  
49 a thin and water-permeable polyamide (PA) rejection layer is supported by a porous  
50 substrate.<sup>3-6</sup> The PA rejection layer exhibits a “ridge-and-valley” surface roughness,<sup>7-9</sup>  
51 and the protuberances of roughness structures extend from a few tens to hundreds of  
52 nm away from the membrane surface.<sup>10, 11</sup> Consequently, some early studies report an  
53 (apparent) thickness in the range of ~ 50 to 300 nm for the rejection layer.<sup>12-15</sup>

54  
55 Advanced microscopic characterization coupled with mechanistic investigation in  
56 recent years has significantly improved our understanding on the structure of the PA  
57 rejection layer.<sup>3, 14, 16-22</sup> For example, transmission electron microscopy (TEM) reveals  
58 the presence of a large fraction of nanovoids inside this layer,<sup>3, 17-19, 23</sup> such that its true  
59 thickness<sup>19, 20, 24-27</sup> would be an order of magnitude lower than the apparent thickness  
60 reported in earlier studies. More recent works<sup>17, 28-30</sup> have shown that the backside of  
61 the PA layer presents numerous honeycomb-like openings. According to Tang and  
62 coworkers,<sup>23, 30-32</sup> both the nanovoids and the honeycomb-like openings originate from  
63 degassing of CO<sub>2</sub> nanobubbles under the combined action of heat and H<sup>+</sup> generated by  
64 the IP reaction. The degassed nanobubbles are encapsulated by the PA film at the  
65 reaction interface to form the nanovoids, and their subsequent escape from porous

66 substrate leaves the honeycomb-like openings at the backside of the PA film.<sup>30</sup> Adding  
67 NaHCO<sub>3</sub>, a CO<sub>2</sub> precursor, to the aqueous MPD solution resulted in larger roughness  
68 protuberances.<sup>31</sup> In contrast, removal of CO<sub>2</sub> from the MPD solution before the IP  
69 reaction eliminated the formation of “ridge-and-valley” surface roughness.<sup>23</sup>

70

71 Despite the apparent importance of these morphological features (the PA film  
72 thickness,<sup>19, 24, 25, 27, 33</sup> its surface roughness,<sup>30, 34-36</sup> and its void fractions,<sup>19, 31, 37-39</sup> etc.)  
73 on the properties of TFC RO membranes, a number of critical issues remain to be  
74 systematically addressed. For example, recent works based microscopic  
75 characterization and tracer tests have provided evidences of pore connectivity within  
76 PA films,<sup>19, 22</sup> yet its underlining mechanism needs to be further investigated. Another  
77 critical question is whether the PA rejection layer is prone to the formation of defects.  
78 Ideally, the free volume pores (~ 0.2 nm in radii<sup>7, 40</sup>) in RO membranes should be small  
79 enough to completely remove viruses and bacteria. Nevertheless, incomplete and often  
80 disparate removal data have been reported for intact membrane elements (e.g., 5 to 7  
81 log removal for the MS2 virus of ~25 nm in size,<sup>41, 42</sup> 3 to ≥4 log removal for a DNA  
82 of ~71 nm in size;<sup>43</sup> and 6.8-6.9 log removal for bacteria surrogates of ~500 nm in size  
83 <sup>44</sup>). In addition to issues related to membrane modules (e.g., failure of glue lines), could  
84 the structure of polyamide film inherently affect the integrity of the membrane?

85

86 Motivated by these questions, the current study aims to establish the structure-  
87 properties relationship and its underlying mechanisms for TFC membranes. The

88 morphological features of four commercial PA membranes were thoroughly  
89 characterized and correlated to their permeability and rejection properties. Additional  
90 tracer filtration tests were designed to reveal the membrane connectivity and defect  
91 location. The results of this study not only provide fundamental insights to the transport  
92 properties of RO membranes but also have critical implications to their synthesis and  
93 environmental applications.

94

## 95 **EXPERIMENTAL**

### 96 *Chemicals and materials*

97 Humic acid (HA) was purchased from Sigma-Aldrich (H16752, technical grade, St.  
98 Louis, MO). N,N-dimethylformamide (DMF), Sodium Chloride (NaCl), Sodium  
99 hypochlorite (NaOCl), Ferric (II) sulfate (FeSO<sub>4</sub>, or Fe(II)) were supplied by Aladdin  
100 (Shanghai, China). Four commercial TFC PA membranes, including a seawater RO  
101 membrane SW30HR, two brackish water RO membranes BW30 and XLE, and a  
102 nanofiltration membrane NF90, were all obtained from Dow Filmtech<sup>TM</sup> (Minneapolis,  
103 MN).<sup>45</sup> All these membranes have a fully aromatic PA chemistry, formed based on the  
104 reaction between MPD and TMC.<sup>46, 47</sup> Compared to SW30HR and BW30, XLE and  
105 NF90 are known as “loose” PA membranes<sup>47</sup> with significantly higher water  
106 permeability but lower NaCl rejection (Table 1). In addition to the commercial TFC PA  
107 membranes, a polysulfone (PSF) ultrafiltration membrane (Aromem Pte. Ltd., Suzhou,  
108 China) was used as a control for the membrane chlorination treatment.

109

110 **Table 1.** Separation properties and structural characteristics of the commercial TFC  
 111 membranes (all data obtained from the current study)  
 112

Membrane		SW30HR <sup>c</sup>	BW30	XLE	NF90
Separation performance at low salinity <sup>a</sup>	Water flux, $J$ (L m <sup>-2</sup> h <sup>-1</sup> )	9.0 ± 0.8	61.4 ± 2.6	101.8 ± 1.4	125.4 ± 1.0
	Salt rejection, $R$ (%)	98.5 ± 0.2	96.9 ± 0.3	92.6 ± 0.5	83.8 ± 0.3
Separation performance at high salinity <sup>b</sup>	Water flux, $J$ (L m <sup>-2</sup> h <sup>-1</sup> )	18.9 ± 1.7	95.4 ± 0.7	120.0 ± 0.9	125.0 ± 1.3
	Salt rejection, $R$ (%)	99.4 ± 0.1	97.4 ± 0.3	92.1 ± 0.5	84.3 ± 0.2
Pores on substrate	Diameter, $d_p$ (nm)	15.2 ± 4.8	9.1 ± 1.9	15.5 ± 3.8	13.1 ± 5.0
	Number density (counts/μm <sup>2</sup> )	439	872	525	932
	Area coverage (%)	11.6	5.9	10.9	14.5
Openings on PA backside	Diameter, $d_o$ (nm)	39.1 ± 12.4	52.9 ± 16.0	51.5 ± 17.1	38.6 ± 10.4
	Number density (counts/μm <sup>2</sup> )	168	106	158	142
	Area coverage (%)	27.1	25.4	47.5	22.9
Nodule size $d_n$ , measured by SEM (nm)		46.7 ± 14.5	60.2 ± 9.5	84.8 ± 27.7	67.1 ± 13.8
$d_n/d_o$ ratio		1.2	1.1	1.6	1.7
Average roughness $R_a$ (nm) <sup>e</sup>		48.8 ± 2.5	36.0 ± 1.4	61.4 ± 1.8	50.6 ± 7.1
Root Mean Square (RMS) roughness $R_q$ (nm) <sup>e</sup>		60.6 ± 3.5	45.6 ± 2.3	77.6 ± 3.5	66.8 ± 9.3
Maximum roughness $R_{max}$ (nm) <sup>e</sup>		425 ± 59	355 ± 38	560 ± 34	540 ± 93
Surface area ratio (SAR), measured by TEM (-) <sup>d</sup>		3.35	3.78	7.05	5.62
SAR, measured by AFM (-) <sup>e</sup>		1.290	1.341	1.329	1.252
$\delta_{app}$ , measured by TEM (nm) <sup>d,f</sup>		129 ± 61	118 ± 58	277 ± 43	159 ± 27
$\delta_{app}$ , measured by AFM (nm) <sup>e,g</sup>		169 ± 11	112 ± 16	204 ± 24	122 ± 8
$\delta_{mr}$ (nm) <sup>d</sup>		17.4 ± 2.0	17.8 ± 1.9	13.9 ± 1.8	11.8 ± 1.7

- 113 a. Brackish water desalination testing conditions: 1.6 MPa, 2000 ppm NaCl, 25 ± 1 °C. Refer to  
 114 Supporting Information S1 for more detailed testing information.  
 115 b. Seawater desalination testing conditions: 5.5 MPa, 32000 ppm NaCl, 25 ± 1 °C. Refer to  
 116 Supporting information S1 for more detailed testing information.  
 117 c. The condition duration was 24 hours for SW30HR and 1 hour for the other 3 membranes.  
 118 d. Based on TEM characterization.  
 119 e. Based on atomic force microscopy (AFM) characterization.  
 120 f. The apparent thickness,  $\delta_{app}$ , was obtained from TEM images by sampling at different locations  
 121 (see Figure S4 in Supporting Information S5).  
 122 g. The apparent thickness,  $\delta_{app}$ , was calculated based on the average height of isolated polyamide  
 123 films by AFM measurements (Figure 2).

124 ***Isolation of PA rejection layer and PSF substrate***

125 To obtain isolated PA rejection layer from a TFC membrane, its PSF substrate was  
126 dissolved by pure DMF.<sup>15, 30</sup> The isolated PA layer was picked up from the DMF  
127 solution by a stainless steel washer, transferred to a silicon wafer, and then further  
128 treated by three cycles of DMF rinsing (3 min) and soaking (10 mins) to remove any  
129 residual PSF. The treated sample was kept in a vacuum dryer for further characterization.

130

131 For the isolation of the PSF substrate, the PA layer of the TFC membrane was removed  
132 by washing with 1% NaOCl solution at pH 12 for 15 hours. The combination of high  
133 hypochlorite concentration and high pH is known to completely degrade the PA layer.<sup>48</sup>

134 At the same time, controlled experiment of NaOCl treatment for the PSF ultrafiltration  
135 membrane under identical conditions did not result in significant changes of its pore  
136 size distribution and pure water permeability (Supporting Information S2). Therefore,  
137 the NaOCl treatment in the current study is expected to remove the PA layer while  
138 preserving the pore structure of the substrate.

139

140 ***Microscopic characterization***

141 The surface morphology and cross-sectional structure of the membranes were  
142 characterized by an ultrahigh-resolution field-emission scanning electron microscope  
143 (FESEM, Hitachi SU8010, Japan). Membrane cross-sections were produced by  
144 fracturing membrane coupons in liquid N<sub>2</sub>. In addition, FESEM characterization was  
145 also performed for PA rejection layers and PSF substrates isolated from the commercial



146 TFC membranes. Prior to FESEM observation, the samples were sputter coated with Pt  
147 at a coating distance of  $\sim 8$  cm with a current of 15 mA. The coating time for membrane  
148 front surface and cross-section was 30 and 45 s, respectively. A shorter coating time of  
149 20 s was used for the backside of the isolated PA layer as well as the front surface of  
150 the isolated PSF substrate to minimize the impact of Pt deposition on the determination  
151 of pore size.

152

153 Cross-sectional TEM images were acquired with a JEM1200EX TEM (acceleration  
154 voltage of 80 kV, JOEL, Tokyo, Japan). Membrane samples were dehydrated gradually  
155 in a series of ethanol/water solution (0%, 25%, 50%, 75%, and 100% ethanol content,  
156 each step for 5 minutes) and subsequently embedded in LR White Resin (London Resin  
157 Company, Reading, UK). The resin-embedded membrane samples were sliced into  
158 ultrathin sections of  $\sim 80$  nm using a Leica EM UC7 (Leica Microsystems, Wetzlar,  
159 Germany).<sup>23, 46</sup> An AFM (ICON, Bruker, Billerica, MA) was used to obtain the  
160 topological images of isolated PA layers of SW30, BW30, XLE, and NF90 (placed on  
161 a silicon wafer) using a tapping mode.<sup>30</sup>

162

### 163 *Tracer filtration tests*

164 Tracer filtration tests (also known as penetration tests in some earlier studies<sup>49</sup>) were  
165 performed for BW30 and NF90 in a customized filtration cell (effective filtration area  
166 of  $\sim 19.6$  cm<sup>2</sup>) to assess the interconnectivity within the PA layer and to determine the  
167 location where separation takes place. HA was chosen as a tracer in this study due to its

168 good solubility and low cost. Its small size ( $\sim 1.1 - 5.4 \text{ nm}^{50}$ ) and soft nature may allow  
169 it to better penetration into pores and defects compared to rigid silver or gold  
170 nanoparticles.<sup>8, 22, 49, 51</sup> For a typical test, a virgin membrane was operated in the “reverse  
171 orientation” with its substrate side exposed to the feed solution (a 5 mg/L HA solution)  
172 under a constant pressure of 0.69 MPa (i.e., 100 psi) for 96 hours. Such a tracer test in  
173 the “reverse orientation” allows HA macromolecules to access the backside of the PA  
174 layer. Additional HA tracer tests were performed in the “normal orientation” to allow  
175 HA macromolecules to access the frontside of the PA layer (see more details in  
176 Supporting Information S8). As a complimentary test to the pressure-driven tests, an  
177 osmotically driven filtration test was also performed using NaCl as draw solution and  
178 Fe (II) as a tracer (Supporting Information S7) with the backside of the PA layer exposed  
179 to the tracer.

180

## 181 **RESULTS AND DISCUSSION**

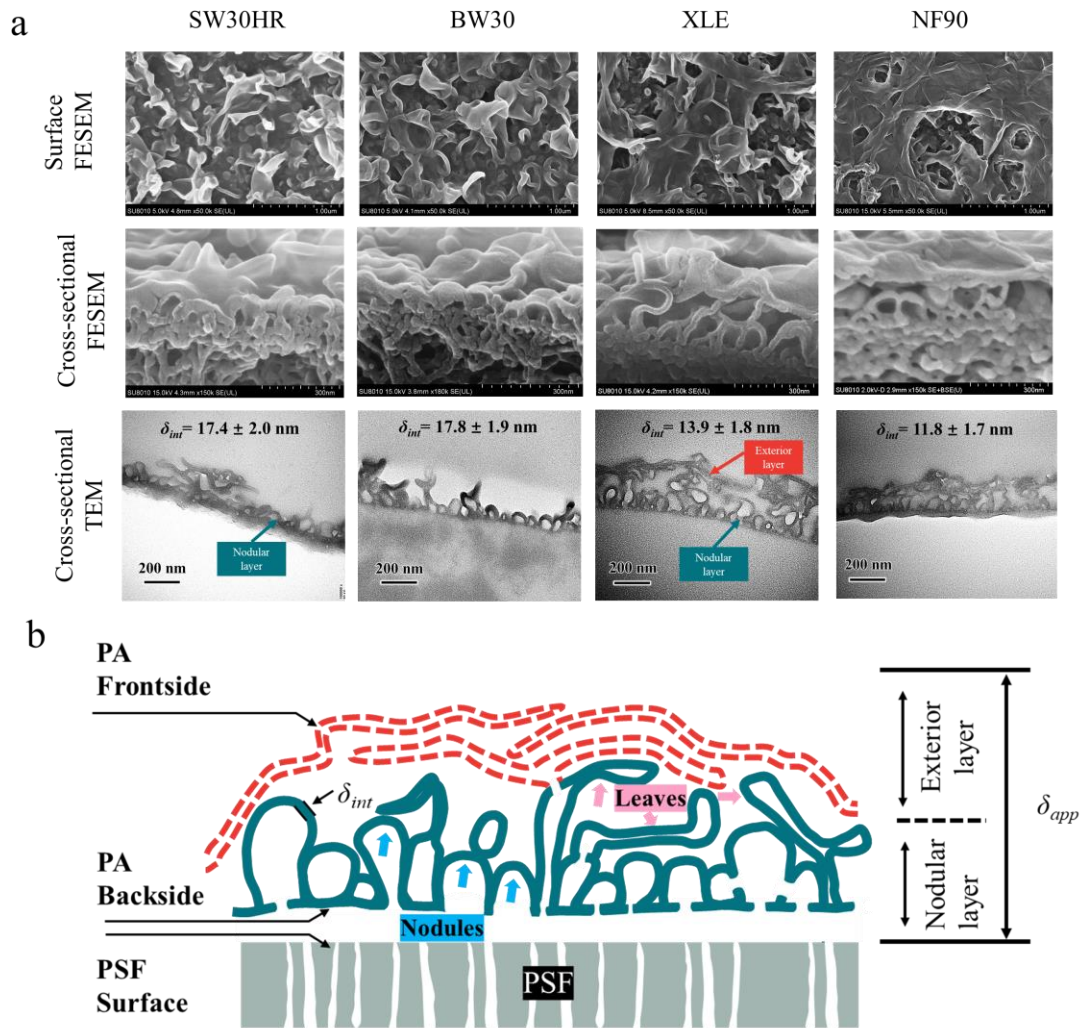
### 182 *Morphology of PA rejection layers*

183 The micrographs of membrane top surfaces and cross-sections are presented in Figure  
184 1a. The surfaces of SW30HR and BW30 have a “ridge-and-valley” appearance that is  
185 characterized by the presence of numerous nodules and leaves.<sup>8, 17, 47 11, 52</sup> Their cross-  
186 sectional FESEM and TEM micrographs show a basal layer of nanovoid-containing  
187 nodular and leave-like features that are densely packed side by side. According to our  
188 previous studies,<sup>23, 30</sup> the nodular features originate from the interfacial degassing of  
189 CO<sub>2</sub> nanobubbles during the IP reaction that are subsequently encapsulated by the PA

190 layer. On the other hand, the dehydration of larger nodules causes them to collapse into  
191 a leaf-like appearance during membrane drying.<sup>30</sup> For these reasons, this basal layer is  
192 termed as the “nodular layer” in the current study.

193

194 XLE and NF90, both considered as loose TFC PA membranes with high permeability  
195 and low NaCl rejection (Table 1),<sup>47</sup> present substantially different PA morphology that  
196 is characterized by a dual-layer structure with a basal nodular layer overlaid by an  
197 additional exterior layer (see the FESEM and TEM cross-sections in Figure 1a). Based  
198 on the FESEM surface micrographs, the exterior layer consists largely interconnected  
199 and relatively flat features (Supporting Information S3). This exterior layer covers  
200 substantial fraction of the surface, leaving some crater-like regions with the underlying  
201 basal nodular features exposed. The exterior layer could not be removed by DMF  
202 washing or long-term filtration operation, confirming that this layer is firmly connected  
203 to the basal layer. Additional analysis (Supporting Information S4 and Table 1) shows  
204 that average size of the nodules ( $d_n$ ) following the order of XLE ( $84.8 \pm 27.7$  nm) >  
205 NF90 ( $67.1 \pm 13.8$  nm) > BW30 ( $60.2 \pm 9.5$  nm) > SW30HR ( $46.7 \pm 14.5$  nm). The  
206 unique dual layer roughness structure of XLE and NF90 (Figure 1b) may be explained  
207 by more favorable CO<sub>2</sub> degassing during their interfacial polymerization. Previous  
208 studies reported the formation of larger nodules due to improved CO<sub>2</sub> degassing.<sup>31, 32</sup>  
209 Indeed, dosing 2-6 wt.% NaHCO<sub>3</sub> (a CO<sub>2</sub> precursor) into the MPD monomer solution  
210 led to the formation of a dual layer structure with similar appearance to those of XLE  
211 and NF90.<sup>31</sup>



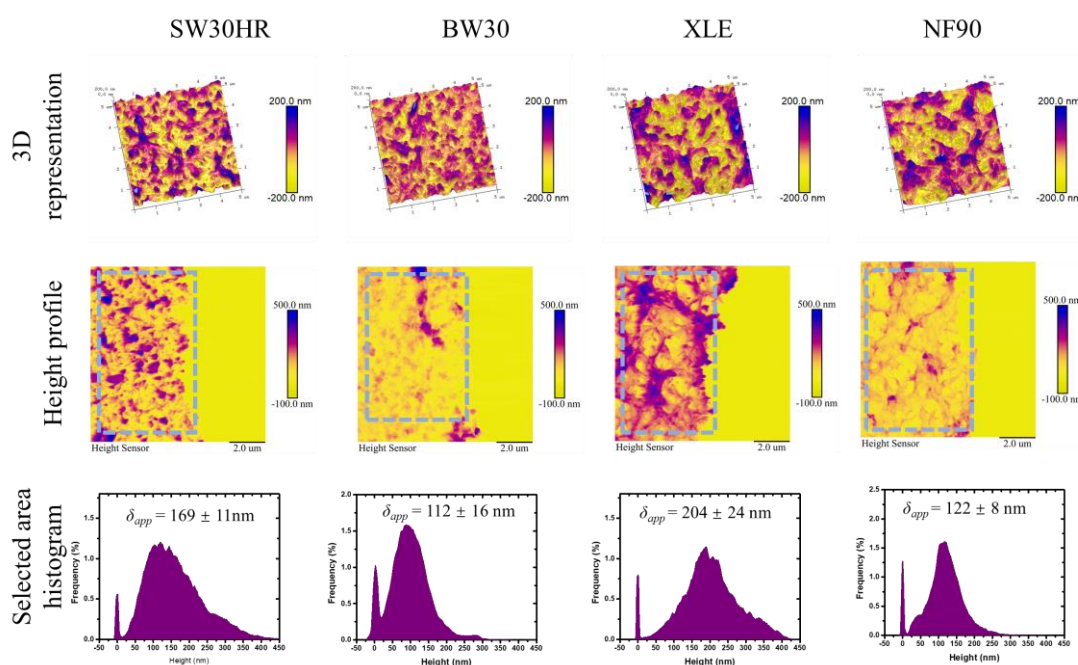
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213 **Figure 1.** (a) The surface FESEM, cross-sectional FESEM, and cross-sectional TEM  
 214 images of SW30HR, BW30, XLE, and NF90 membranes; (b) The schematic illustration  
 215 of a PA layer featuring a dual layer structure, with a basal nodular layer composed of  
 216 tightly packed nodules (represented by the green solid lines) and an exterior layer  
 217 composed of interconnected flat features (represented by the red dotted lines). The dual  
 218 layer structure was observed in XLE and NF90 membrane, but not in BW30 and  
 219 SW30HR membrane.

220

221 *Apparent thickness, intrinsic thickness, and effective filtration area of PA layers*

222 The apparent thickness ( $\delta_{app}$ ) of the PA layers, measured from the cross-sectional TEM  
 223 images (Figure 1a and Figure S4a in Supporting Information S5), followed the order of  
 224 XLE ( $277 \pm 43$  nm) > NF90 ( $159 \pm 27$  nm) > SW30HR ( $129 \pm 61$  nm) > BW30 ( $118 \pm$   
 225  $58$  nm). AFM height analysis provides an alternative measurement of  $\delta_{app}$  based on the  
 226 height histogram (Figure 2). Due to the existence of the additional exterior layer, the  
 227  $\delta_{app}$  values of XLE ( $204 \pm 24$  nm) and NF90 ( $122 \pm 8$  nm) were approximately double  
 228 of their average nodular size. SW30HR and BW30 had apparent thicknesses of  $169 \pm$   
 229  $11$  nm and  $112 \pm 16$  nm, respectively, which were significantly larger than the average  
 230 nodular size partially due to the presence of leaf-like features. Furthermore, the  
 231 additional surface coating layer of polyvinyl alcohol (PVA) of these two membranes<sup>46</sup>,  
 232 <sup>47</sup> also result in over-estimations of  $\delta_{app}$  values by the AFM method, since AFM is not  
 233 able to differentiate PVA from PA.



234

235 **Figure 2.** The AFM 3D representation, height profile, and selected area histograms of  
 236 the isolated PA layers of SW30HR, BW30, XLE, and NF90 membranes.

237

238 The intrinsic PA thickness ( $\delta_{int}$ , Figure 1b and Table 1) was estimated from the cross-  
239 sectional TEM images (Figure 1a and Figure S4b in Supporting Information S5). For  
240 any given membrane, its nodules had relatively uniform intrinsic layer thicknesses in  
241 the range of 10 - 20 nm, which is in good agreement with the existing literature.<sup>17, 24, 35,</sup>  
242 <sup>53</sup> Similar intrinsic thickness was also reported by Song et al.<sup>19</sup> for roughness nodules  
243 (“domes”). Using a quantitative electron tomography, these authors also found that  
244 “dimpled” and “clustered” crumples (which are possibly related to collapsed leaf-like  
245 features and the secondary layer reported in this study, respectively) could have greater  
246 film thickness. In the current study, XLE ( $13.9 \pm 1.8$  nm) and NF90 ( $11.8 \pm 1.7$  nm) had  
247 thinner  $\delta_{int}$  compared to SW30HR ( $\delta_{int} = 17.4 \pm 2.0$  nm) and BW30 ( $17.8 \pm 1.9$  nm),  
248 which partially explains the greater water permeability of the loose XLE and NF90  
249 membranes (Table 1). This finding also underpins the recent attempts of making thinner  
250 PA rejection layer in order to achieve higher membrane water permeability.<sup>23, 24, 33, 35, 54,</sup>

251 <sup>55</sup>

252

253 Membrane water permeability is also greatly affected by the effective PA surface area,<sup>23,</sup>  
254 <sup>53</sup> which is in turn affected by the membrane surface roughness features. The average  
255 surface roughness values ( $R_a$ , see Figure 2 and Table 1) of the loose membranes XLE  
256 and NF90 were  $61.4 \pm 1.8$  nm and  $50.6 \pm 7.1$  nm, respectively. Those of SW30HR and  
257 BW30 were  $48.8 \pm 2.5$  nm and BW30  $36.0 \pm 1.4$  nm, respectively. The SAR, i.e., the  
258 ratio of the three-dimensional (3D) surface area measured by AFM over its projected

259 plan area, had the following order (Table 1): BW30 (1.341) > XLE (1.329) > SW30HR  
260 (1.290) > NF90 (1.252). In the current study, neither  $R_a$  or SAR measured by AFM  
261 seems to provide satisfactory explanation to the ten-fold higher flux of the loose  
262 membranes (XLE and NF90) compared to SW30HR under the low salinity testing  
263 conditions (Table 1). This is due to the inherent limitation of AFM measurement that  
264 only the front (exposed) area can be accessed by its tip of finite size.<sup>31</sup>

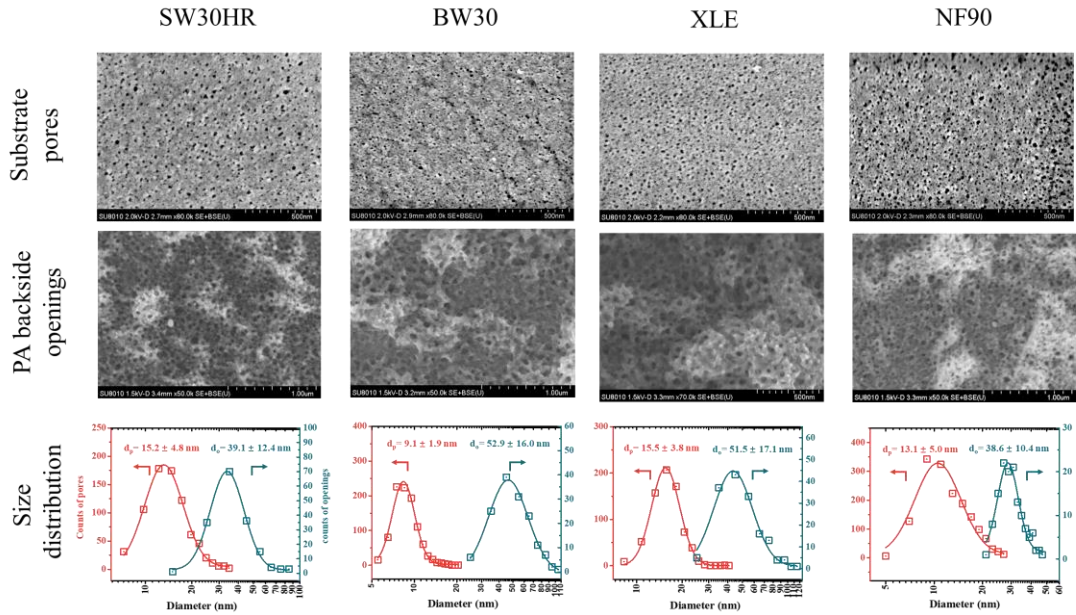
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266 In comparison to the AFM data, the SAR values obtained from TEM cross-sections  
267 (Supporting Information S6 and Table 1) were substantially larger. For example, the  
268 SAR value of SW30HR estimated from the 2D TEM image analysis was 3.35, which  
269 was much larger than the AFM-based SAR of 1.290. A similar trend has been observed  
270 by Culp et al.<sup>21</sup>. Furthermore, the SAR value obtained in this study agrees reasonably  
271 well with the value obtained from the 3D TEM tomography<sup>21</sup> (3.35 vs. 3.48 for  
272 SW30HR), suggesting that the 2D image analysis described in this study could offer a  
273 simpler alternative to the more sophisticated 3D approach. Compared to SW30HR and  
274 BW30, loose membranes XLE and NF90 had much larger SAR values (Table 1), which  
275 is attributed to the additional area contributed by the exterior layer. For example, the  
276 SAR values of XLE estimated from its TEM cross-section (Supporting Information S6)  
277 was as large as 7.05. Our result is also consistent with the 3D TEM measurements by  
278 Song et al.<sup>19</sup>, who reported that the localized SAR varied from 3.5 to 15.7 for a  
279 polyamide film synthesized on a sacrificial layer. Compared to AFM measurements, the  
280 TEM-based SAR provides better explanation to the significantly higher water

281 permeability of the loose PA membranes.

282

283 ***Hierarchical pore structure and interconnectivity within the PA layer***



284

285 **Figure 3.** FESEM micrographs and pore size distributions of the PSF substrates and

286 the backside openings of the PA rejection layer. To isolate the PSF substrate, the PA

287 layer of the TFC membrane was removed using 1% NaOCl solution at pH 12. On the

288 other hand, the PA rejection layer was isolated by dissolving the PSF substrate in DMF.

289 For the size distribution curves, the red and green curves correspond to the pore sizes

290 on the polysulfone surface and polyamide backside, respectively.

291

292 Figure 3 shows FESEM micrographs of the membrane substrates and the backsides of

293 PA layers. The backside of all the membranes features honeycomb-like openings, with

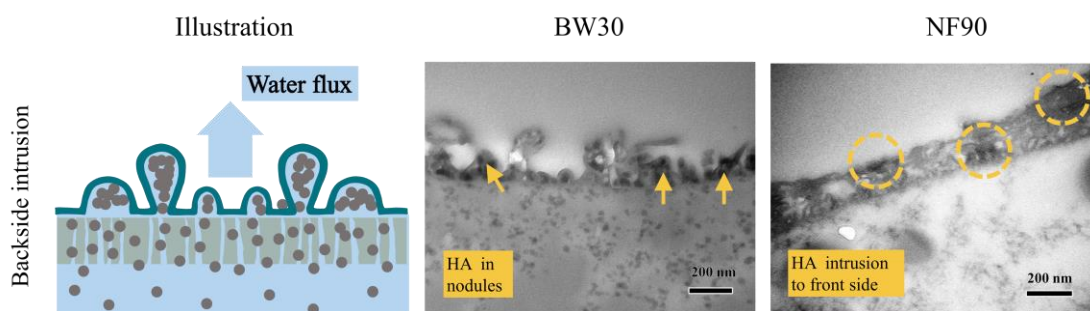
294 the opening density in the range of 100 to 200 counts/ $\mu\text{m}^2$  and the average opening

295 diameter ( $d_o$ ) in the range of 39 to 53 nm (Figure 3 and Table 1). These data agree

296 reasonably well with previous studies.<sup>17, 28, 30</sup> Based on FESEM and TEM cross-



297 sectional micrographs (Figure 1) and previous studies<sup>22, 30</sup>, we hypothesize that the  
298 openings in the backside are directly connected to the nanovoids in the PA layer.  
299 Meanwhile, the density of the substrate pores (in the range of 439 to 932 counts/ $\mu\text{m}^2$ )  
300 was 2.6 - 8.2 times of that of the PA backside openings for the corresponding membrane,  
301 suggesting that each opening may be connected to multiple pores in the substrate.



302  
303

304 **Figure 4.** The illustration of HA tracer test from the backside of the PA layer (left panel)  
305 and TEM micrographs of BW30 and NF90 after the HA tracer test (right panel).

306

307 In order to further study the connectivity within the hierarchical pore structure, we  
308 performed tracer filtration tests for BW30 and NF90 from the substrate side using HA  
309 as a molecular tracer (Figure 4). The macromolecular nature of HA (approximate size  
310 range from 1.1 – 5.4 nm<sup>50</sup>) allows it to penetrate through pores and defects but not an  
311 intact PA film. For both membranes, HA was able to reach all the way to the nanovoids  
312 within the PA layer. Our observation is consistent with Li et al.<sup>22</sup> who reported the  
313 penetration of 1-nm gold nanoparticles into the interior of polyamide nodules. An  
314 additional test was also performed using FeSO<sub>4</sub> as a tracer on the backside of a PA layer  
315 isolated from BW30 using an osmotically driven water permeation, which led to the  
316 accumulation of iron hydroxide deposition within the nodular structures (Supporting

317 Information S7). Earlier studies believe that separation of solutes occurs at a “dense  
318 base” of the PA layer.<sup>8,9</sup> Nevertheless, the observation of honeycomb-like openings at  
319 its backside (Figure 3) and the ability of both HA and FeSO<sub>4</sub> tracers to access nanovoids  
320 in the roughness nodules (Figure 4 and Supporting Information S7) provide strong  
321 evidence that the actual separation takes place at the frontside of the PA layer instead  
322 of its backside.

323

324 Tracer tests by HA from the frontside was also performed for BW30 and NF90  
325 (Supporting Information S8). For the tight BW30 membrane, HA deposited above the  
326 nodules/leaves of the PA layer, whereas the nodular voids within the PA layer and the  
327 pores in the substrate were largely free of HA. For the loose NF90 membrane, large  
328 amount of HA particles deposited in the gap between its exterior layer and nodular layer  
329 of the roughness structure. Moreover, some HA particles penetrated through the nodules  
330 and deposited within their nanovoids. A tiny fraction of HA even reached the pores of  
331 the substrate (marked by the red arrows in Figure S8). We further compared the  
332 rejection of HA by NF90 and BW30. While HA was found in the permeate of NF90, it  
333 was below the detection limit in the permeate of BW30 (Figure S9, Supporting  
334 Information S8). These results suggest the possible presence of defects in the PA layer  
335 of NF90. On the other hand, the better integrity of BW30 may be attributed to a more  
336 intact polyamide film and/or its PVA surface coating<sup>46</sup>. Additional high resolution  
337 FESEM micrographs of NF90 (Figure S10) reveal the frequent existence of nanosized  
338 holes of 10-50 nm in size on the nodular walls of its PA layer. These holes can provide

339 interconnectivity between adjacent nodules.<sup>19</sup> If not completely enclosed during the  
340 film growth, they could also potentially lead to defects through the PA layer. The  
341 possible existence of defects within the PA rejection layer, in addition to integrity of  
342 glue lines in membrane modules, may explain the incomplete removal of viruses whose  
343 physical sizes (e.g., of several tens of nm) are several orders of magnitude larger than  
344 the pore size of RO membranes.<sup>42</sup> It also provides insights to the observation that a  
345 series of RO membranes, even though having similar PA network pore-volumes, can  
346 exhibit significantly different rejection rates to small neutral solutes.<sup>56</sup> Furthermore, it  
347 is consistent with the observation that membrane rejection can be significantly  
348 improved under mild/initial fouling by macromolecules thanks to the partial sealing of  
349 defects.<sup>57</sup>

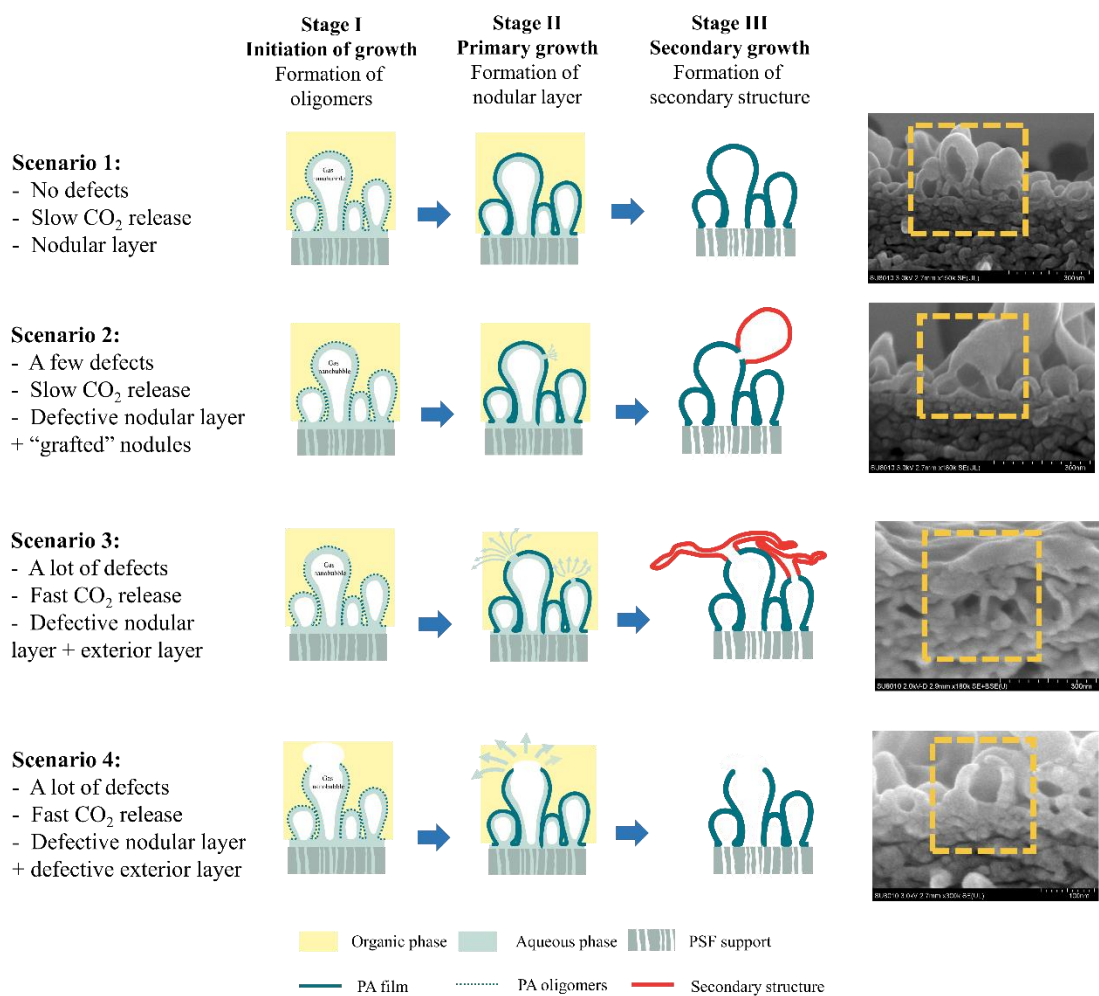
350

351 *Mechanistic insights of roughness structures, inter-connectivity, and defects*  
352 *formation of PA layers*

353 Figure 5 presents a schematic diagram to illustrate the mechanisms involved in forming  
354 the roughness structures as well as defects of PA layers. During the IP reaction, MPD  
355 and TMC react rapidly to form oligomers (Stage I) and eventually a nascent cross-  
356 linked network of polyamide (Stage II). CO<sub>2</sub> nanobubbles released during IP and  
357 confined between the substrate and the nascent PA film lead to the formation of  
358 “balloon-like” nodular roughness features,<sup>23, 30, 31</sup> while the thickness of the PA film can  
359 be strongly affected by the reaction recipe (e.g., monomer concentrations<sup>26, 27, 33</sup>). The  
360 release of gas nanobubbles can also potentially lead to convection of monomer

361 solutions, which can further influence the integrity as well as the final shape of the PA  
362 layer (Stage III). For example, when the PA layer completely encapsulates CO<sub>2</sub>  
363 nanobubbles, a defect-free dense RO membrane is expected (Figure 5, Scenario 1).  
364 However, in some cases, the gas bubble may start to escape from a nascent PA nodule  
365 before it completely encapsulates the gas bubble. This escape of the bubble drives the  
366 convection of the MPD solution towards the TMC solution, which promotes the  
367 formation of secondary structures such as a secondary nodule (or “grafted” nodule) that  
368 is well connected to the primary nodule (Scenario 2). Under conditions where rigorous  
369 degassing prevails, the escape of the gas bubble can be so forceful that the reaction  
370 frontier assumes a shape similar to a pressurized water jet instead of the common  
371 balloon-like shape, which is responsible for the formation of the exterior layer with a  
372 large aspect ratio of length/width (Scenario 3). Extremely forceful degassing and/or the  
373 depletion of MPD monomers may further result in the formation of defects such as the  
374 incomplete enclosure of the primary nodules (Scenario 4). In reality, a combination of  
375 the various scenarios can be applicable to the same membrane (e.g., the observation of  
376 all the four scenarios for NF90), which can be attributed to the highly heterogeneous IP  
377 reaction conditions at the nanoscale.

378



379

380 **Figure 5.** Formation mechanisms of the nodular layer and the exterior layer.

381

382 In the current study, we performed XPS elemental composition analysis of both the  
 383 frontside and backside of PA layers isolated from NF90 and XLE (Table S1). The O/N  
 384 ratios of the frontside of XLE (1.50) and NF90 (1.36) were much higher than the  
 385 respective values of the backside (1.07 for XLE and 1.01 for NF90), revealing that the  
 386 cross-linking degree of the exterior layer was much lower compared to that of the  
 387 nodular layer.<sup>46</sup> In contrast, tight RO membranes have been reported to have more  
 388 uniform chemical compositions across their PA layers.<sup>58</sup> The large difference in the O/N  
 389 ratios between of front and back sides for the loose membranes XLE and NF90 supports

390 our explanation of the formation of exterior layer due to gas escape from the primary  
391 nodules. That is, the forced convection of the MPD solution into the TMC solution  
392 would induce a more TMC-rich and MPD-lean environment, which would in turn form  
393 less crosslinked<sup>6, 13</sup> secondary PA structure in good agreement to our experimental  
394 observation. Interestingly, Table 1 shows that the  $d_n/d_o$  ratios, representing the relative  
395 size of nodules over that of backside pore openings, of the loose membranes XLE (1.6)  
396 and NF90 (1.7) were significantly greater than those of the tight membranes SW30HR  
397 (1.2) and BW30 (1.1). This observation is consistent with the possibly more intensive  
398 degassing during the formation of the loose membranes.

399

Table 2. Correlation coefficients of key measured parameters of commercial RO membranes.<sup>a</sup>

		<i>J</i> (Brackish)	<i>R</i> (Brackish)	Pore diameter	Pore density	Pore coverage	Opening diameter	Opening density	Opening coverage	<i>d<sub>n</sub></i>	<i>R<sub>a</sub></i>	SAR (AFM)	SAR (TEM)	$\delta_{app}$	$\delta_{int}$
<i>J</i> (Brackish)		1													
<i>R</i> (Brackish)		<b>-0.88</b>	1												
Substrate Pore	Diameter	-0.04	-0.09	1											
	Density	<b>0.58</b>	<b>-0.58</b>	<b>-0.76</b>	1										
	Coverage	0.37	<b>-0.69</b>	<b>0.68</b>	-0.08	1									
Opening	Diameter	0.11	0.36	-0.42	0.05	<b>-0.80</b>	1								
	Density	-0.19	-0.01	<b>0.98</b>	<b>-0.78</b>	<b>0.68</b>	<b>-0.53</b>	1							
	Coverage	0.21	0.11	<b>0.54</b>	<b>-0.56</b>	-0.03	<b>0.53</b>	0.40	1						
Nodule size, <i>d<sub>n</sub></i>		<b>0.79</b>	-0.47	0.20	0.08	0.11	<b>0.51</b>	0.01	<b>0.76</b>	1					
<i>R<sub>a</sub></i>		0.38	-0.35	<b>0.88</b>	<b>-0.52</b>	<b>0.63</b>	-0.13	<b>0.78</b>	<b>0.75</b>	<b>0.64</b>	1				
SAR (AFM)		-0.25	<b>0.67</b>	-0.33	-0.22	<b>-0.88</b>	<b>0.93</b>	-0.39	<b>0.50</b>	0.24	-0.20	1			
SAR (TEM)		<b>0.81</b>	<b>-0.60</b>	0.42	0.00	0.39	0.24	0.24	<b>0.72</b>	<b>0.96</b>	<b>0.80</b>	-0.02	1		
$\delta_{app}$		-0.11	0.23	<b>0.85</b>	<b>-0.87</b>	0.23	0.12	<b>0.78</b>	<b>0.87</b>	0.42	<b>0.83</b>	0.22	0.49	1	
$\delta_{int}$		<b>-0.88</b>	<b>0.95</b>	-0.35	-0.32	<b>-0.77</b>	0.31	-0.24	-0.17	<b>-0.63</b>	<b>-0.62</b>	<b>0.60</b>	<b>-0.78</b>	-0.09	1

a. The correlation coefficient  $R^2$  was obtained using Microsoft Excel software. The correlation is classified in strongly correlated ( $|R^2| \geq 0.8$ ), well correlated ( $0.8 > |R^2| \geq 0.5$ ), weakly correlated ( $0.5 > |R^2| \geq 0.3$ ), and no obvious correlation ( $|R^2| < 0.3$ ). Bold font is used to indicate cases of strongly or well correlated cases.

397 ***Structure-property relationship***

398 Table 2 shows the correlation among various parameters for the four TFC membranes.  
399 The salt rejection was strongly negatively correlated to the membrane water flux, which  
400 is expected based on the well-known upper bound theory describing the tradeoff  
401 relationship between membrane selectivity and permeance.<sup>6, 59, 60</sup> Among all the  
402 parameters, the intrinsic PA thickness had the strongest impact on the membrane  
403 separation properties, with a thinner  $\delta_{int}$  resulting in a greater flux ( $R^2 = -0.88$ ) but  
404 reduced NaCl rejection ( $R^2 = 0.95$ ). In addition, greater effective filtration area (as  
405 characterized by the TEM-based SAR) also led to enhanced flux ( $R^2 = 0.81$ ) and poorer  
406 NaCl rejection ( $R^2 = -0.60$ ). The negative correlation between TEM-based SAR and  
407 NaCl rejection was likely due to the greater presence of defects in membranes  
408 containing extensive exterior layers. Compared to the intrinsic PA thickness and the  
409 TEM-based SAR, the more widely used conventional parameters including the  
410 apparent PA thickness and the AFM-based SAR were far less useful for the prediction  
411 of membrane separation properties. In the current study, the flux and rejection of the  
412 membranes also appeared to be correlated to the pore number density and coverage (i.e.,  
413 surface porosity) of the substrate and the average nodular size. Our correlation analysis  
414 supports the earlier studies<sup>30, 31</sup> reporting the positive role of larger nodular size and  
415 greater surface roughness in enhancing membrane permeability by offering greater  
416 effective filtration area (see their strong correlation with TEM-based SAR in Table 2).  
417 The important role of the substrate revealed in this analysis also underpins the increased  
418 interest<sup>61-63</sup> in the optimization of the substrate pore structure. Based on the



419 observations in this study, the higher water permeability of the loose XLE and NF90  
420 membranes can be mainly attributed to their thinner intrinsic PA thickness and larger  
421 effective filtration area, both resulting in reduced hydraulic resistance. On the other  
422 hand, the formation of a more defective PA film and the reduced crosslinking degree  
423 (e.g., for the exterior layer) are likely responsible for the reduced salt rejection.  
424 Despite the insights revealed by the correlation analysis, the readers need to be mindful  
425 that such analysis was only based on a limited collection of commercial membranes (4  
426 types) with proprietary recipes. Furthermore, these commercial membranes were  
427 provided in their dry state, which could cause some deformation of morphological  
428 features.<sup>30</sup> Future studies need to expand the correlation analysis to a wider range of  
429 membranes, including samples synthesized under controlled synthesis conditions and  
430 without drying history.

431

### 432 **Implications and perspectives**

433 The current study suggests that the membrane transport properties are governed by a  
434 selective PA skin of 10-20 nm in intrinsic thickness. On the other hand, the backside of  
435 the PA rejection layer is porous and does not appear to contribute to solute rejection.  
436 The new mechanistic insights gained with respect to the intrinsic structure of the  
437 discriminative PA layer, its connectivity to the substrate pores and the structure-  
438 properties correlation facilitate improved understanding of transport phenomena in RO  
439 membranes. For example, the inherent tendency to form nanosized defects as a result  
440 of interfacial degassing during membrane formation has major implications to the

441 removal of pathogens and micropollutants, both of which are of critical concerns for  
442 water reuse applications.<sup>2</sup> In this respect, additional studies of in-depth material  
443 characterization and multi-scale simulations are needed. For example, RO membranes  
444 are traditionally modelled by the solution-diffusion theory (which assumes diffusion is  
445 the sole transport mechanism). Although some studies have included additional  
446 convective transport of solutes across RO membranes,<sup>64-66</sup> the potential role of defects  
447 on their transport behavior need to be refined based on the latest advances in membrane  
448 characterization. The respective roles of the nodular layer and the exterior layer need to  
449 be further resolved with respect to the transport of both water and solutes. Better  
450 characterization methods are also needed for studying the nanosized defects and voids  
451 within the PA layer as well as for the accurate determination of the effective membrane  
452 transport area. Furthermore, some existing characterization results may need to be  
453 reassessed to determine whether their interpretation could be affected by the presence  
454 of these nanosized features (e.g., pore size measurement of RO membranes based on  
455 the rejection of sugar molecules<sup>67, 68</sup> and position annihilation lifetime spectroscopy<sup>56</sup>).

456

457 The current study also implies the critical importance of optimizing of the ultrathin  
458 selective PA skin. A key challenge would be to prepare a thin yet defect-free PA skin  
459 with large SAR in order to overcome the upper-bound tradeoff<sup>6, 59, 60</sup> of PA membranes.  
460 This in turn calls for the careful optimization of the reaction recipes and the interfacial  
461 degassing conditions to promote the growth of thin polyamide films with larger  
462 filtration area and to avoid extensive defects formation at the same time. One highly

463 promising method is the including of an interlayer<sup>69-71</sup> between the PA and the substrate  
464 layers, which enhances water permeability and minimizes defects formation  
465 simultaneously.

466

### 467 **Supporting Information**

468 The Supporting Information is available free of charge on the ACS Publications website  
469 at DOI:

470 S1. Membrane performance tests; S2. The impact of NaOCl treatment on the PSF  
471 substrate; S3. FESEM micrographs of the exterior features of XLE and NF90; S4.  
472 Determination of nodule size; S5. Determination of the apparent and intrinsic  
473 thicknesses by TEM; S6. Determination of SAR value using TEM micrographs;  
474 S7. Fe(II) tracer filtration test from the backside of BW30; S8. HA tracer filtration  
475 tests from the frontside of BW30 and NF90; S9. FESEM observation of the  
476 defective nodular layer of NF90; S10. XPS elemental composition of the frontside  
477 and backside of isolated PA layers; S11. Effect of DMF treatment on PA  
478 morphology

479

### 480 **Acknowledgement**

481 This research was funded by the National Key R&D Program of China (No.  
482 2017YFC0403903), the National Natural Science Foundation of China (No. 21706231),  
483 and the Seed Funding for Strategic Interdisciplinary Research Scheme, the University  
484 of Hong Kong. Aromem Pte. Ltd. (Suzhou, China) is kindly acknowledged for  
485 providing free PSF substrate samples.

487 **References**

488

- 489 1. Elimelech, M.; Phillip, W. A., The future of seawater desalination: Energy,  
490 technology, and the environment. *Science* **2011**, *333*, (6043), 712.
- 491 2. Tang, C. Y.; Yang, Z.; Guo, H.; Wen, J. J.; Nghiem, L. D.; Cornelissen, E., Potable  
492 Water Reuse through Advanced Membrane Technology. *Environmental Science &*  
493 *Technology* **2018**, *52*, (18), 10215-10223.
- 494 3. Ridgway, H. F.; Orbell, J.; Gray, S., Molecular simulations of polyamide membrane  
495 materials used in desalination and water reuse applications: Recent developments and  
496 future prospects. *J. Membr. Sci.* **2017**, *524*, 436-448.
- 497 4. Cadotte, J. E., Evolution of composite reverse osmosis membranes. In *Materials*  
498 *Science of Synthetic Membranes*, American Chemical Society: 1985; Vol. 269, pp 273-  
499 294.
- 500 5. Petersen, R. J., Composite reverse osmosis and nanofiltration membranes. *J.*  
501 *Membr. Sci.* **1993**, *83*, (1), 81-150.
- 502 6. Yang, Z.; Guo, H.; Tang, C. Y., The upper bound of thin-film composite (TFC)  
503 polyamide membranes for desalination. *Journal of Membrane Science* **2019**, *590*,  
504 117297.
- 505 7. Kim, S. H.; Kwak, S.-Y.; Suzuki, T., Positron annihilation spectroscopic evidence  
506 to demonstrate the flux-enhancement mechanism in morphology-controlled Thin-Film-  
507 Composite (TFC) membrane. *Environ. Sci. Technol.* **2005**, *39*, (6), 1764-1770.
- 508 8. Pacheco, F. A.; Pinnau, I.; Reinhard, M.; Leckie, J. O., Characterization of isolated  
509 polyamide thin films of RO and NF membranes using novel TEM techniques. *J. Membr.*  
510 *Sci.* **2010**, *358*, (1), 51-59.
- 511 9. Freger, V., Nanoscale heterogeneity of polyamide membranes formed by interfacial  
512 polymerization. *Langmuir* **2003**, *19*, (11), 4791-4797.
- 513 10. Kurihara, M.; Hanakawa, M., Mega-ton water system: Japanese national research  
514 and development project on seawater desalination and wastewater reclamation.  
515 *Desalination* **2013**, *308*, 131-137.
- 516 11. Jeong, B.-H.; Hoek, E. M. V.; Yan, Y.; Subramani, A.; Huang, X.; Hurwitz, G.;  
517 Ghosh, A. K.; Jawor, A., Interfacial polymerization of thin film nanocomposites: A new  
518 concept for reverse osmosis membranes. *J. Membr. Sci.* **2007**, *294*, (1), 1-7.
- 519 12. Pacheco, F. A., Microscopic characterization of the nanostructure of polyamide thin  
520 films in reverse osmosis and nanofiltration membranes. *Ph.D. Thesis. Stanford*  
521 *University* **December 2011**.
- 522 13. Freger, V., Kinetics of film formation by interfacial polycondensation. *Langmuir*  
523 **2005**, *21*, (5), 1884-1894.
- 524 14. Freger, V., Swelling and morphology of the skin layer of polyamide composite  
525 membranes: An atomic force microscopy study. *Environ. Sci. Technol.* **2004**, *38*, (11),  
526 3168-3175.
- 527 15. Lin, L.; Feng, C.; Lopez, R.; Coronell, O., Identifying facile and accurate methods  
528 to measure the thickness of the active layers of thin-film composite membranes – A

529 comparison of seven characterization techniques. *J. Membr. Sci.* **2016**, *498*, 167-179.

530 16. Pacheco, F.; Sougrat, R.; Reinhard, M.; Leckie, J. O.; Pinnau, I., 3D visualization  
531 of the internal nanostructure of polyamide thin films in RO membranes. *J. Membr. Sci.*  
532 **2016**, *501*, 33-44.

533 17. Yan, H.; Miao, X.; Xu, J.; Pan, G.; Zhang, Y.; Shi, Y.; Guo, M.; Liu, Y., The porous  
534 structure of the fully-aromatic polyamide film in reverse osmosis membranes. *J. Membr.*  
535 *Sci.* **2015**, *475*, 504-510.

536 18. Lin, L.; Lopez, R.; Ramon, G. Z.; Coronell, O., Investigating the void structure of  
537 the polyamide active layers of thin-film composite membranes. *J. Membr. Sci.* **2016**,  
538 *497*, 365-376.

539 19. Song, X.; Smith, J. W.; Kim, J.; Zaluzec, N. J.; Chen, W.; An, H.; Dennison, J. M.;  
540 Cahill, D. G.; Kulzick, M. A.; Chen, Q., Unraveling the Morphology–Function  
541 Relationships of Polyamide Membranes Using Quantitative Electron Tomography. *ACS*  
542 *Applied Materials & Interfaces* **2019**, *11*, (8), 8517-8526.

543 20. Kłosowski, M. M.; McGilvery, C. M.; Li, Y.; Abellan, P.; Ramasse, Q.; Cabral, J.  
544 T.; Livingston, A. G.; Porter, A. E., Micro-to nano-scale characterisation of polyamide  
545 structures of the SW30HR RO membrane using advanced electron microscopy and  
546 stain tracers. *Journal of Membrane Science* **2016**, *520*, 465-476.

547 21. Culp, T. E.; Shen, Y.-x.; Geitner, M.; Paul, M.; Roy, A.; Behr, M. J.; Rosenberg, S.;  
548 Gu, J.; Kumar, M.; Gomez, E. D., Electron tomography reveals details of the internal  
549 microstructure of desalination membranes. *Proceedings of the National Academy of*  
550 *Sciences* **2018**, *115*, (35), 8694-8699.

551 22. Li, Y.; Kłosowski, M. M.; McGilvery, C. M.; Porter, A. E.; Livingston, A. G.;  
552 Cabral, J. T., Probing flow activity in polyamide layer of reverse osmosis membrane  
553 with nanoparticle tracers. *Journal of Membrane Science* **2017**, *534*, 9-17.

554 23. Ma, X.-H.; Yao, Z.-K.; Yang, Z.; Guo, H.; Xu, Z.-L.; Tang, C. Y.; Elimelech, M.,  
555 Nanofoaming of polyamide desalination membranes to tune permeability and  
556 selectivity. *Environmental Science & Technology Letters* **2018**, *5*, (2), 123-130.

557 24. Kong, C.; Kanezashi, M.; Yamomoto, T.; Shintani, T.; Tsuru, T., Controlled  
558 synthesis of high performance polyamide membrane with thin dense layer for water  
559 desalination. *J. Membr. Sci.* **2010**, *362*, (1), 76-80.

560 25. Yan, W.; Shi, M.; Wang, Z.; Zhao, S.; Wang, J., Confined growth of skin layer for  
561 high performance reverse osmosis membrane. *J. Membr. Sci.* **2019**, *585*, 208-217.

562 26. Karan, S.; Jiang, Z.; Livingston, A. G., Sub-10 nm polyamide nanofilms with  
563 ultrafast solvent transport for molecular separation. *Science* **2015**, *348*, (6241), 1347-  
564 1351.

565 27. Jiang, Z.; Karan, S.; Livingston, A. G., Water transport through ultrathin polyamide  
566 nanofilms used for reverse osmosis. *Adv. Mater.* **2018**, *30*, (15), 1705973.

567 28. Lee, J.; Hill, A.; Kentish, S., Formation of a thick aromatic polyamide membrane  
568 by interfacial polymerisation. *Sep. Purif. Technol.* **2013**, *104*, 276-283.

569 29. Shen, H.; Wang, S.; Xu, H.; Zhou, Y.; Gao, C., Preparation of polyamide thin film  
570 nanocomposite membranes containing silica nanoparticles via an in-situ polymerization  
571 of SiCl<sub>4</sub> in organic solution. *J. Membr. Sci.* **2018**, *565*, 145-156.

572 30. Song, X.; Gan, B.; Yang, Z.; Tang, C. Y.; Gao, C., Confined nanobubbles shape the

573 surface roughness structures of thin film composite polyamide desalination membranes.  
574 *Journal of Membrane Science* **2019**, *582*, 342-349.

575 31. Ma, X.; Yang, Z.; Yao, Z.; Guo, H.; Xu, Z.; Tang, C. Y., Tuning roughness features  
576 of thin film composite polyamide membranes for simultaneously enhanced  
577 permeability, selectivity and anti-fouling performance. *Journal of Colloid and Interface*  
578 *Science* **2019**, *540*, 382-388.

579 32. Peng, L. E.; Yao, Z.; Liu, X.; Deng, B.; Guo, H.; Tang, C. Y., Tailoring polyamide  
580 rejection layer with aqueous carbonate chemistry for enhanced membrane separation:  
581 Mechanistic insights, chemistry-structure-property relationship, and environmental  
582 implications. *Environ. Sci. Technol.* **2019**, *53*, (16), 9764-9770.

583 33. Song, X.; Qi, S.; Tang, C. Y.; Gao, C., Ultra-thin, multi-layered polyamide  
584 membranes: Synthesis and characterization. *Journal of Membrane Science* **2017**, *540*,  
585 10-18.

586 34. Xu, J.; Yan, H.; Zhang, Y.; Pan, G.; Liu, Y., The morphology of fully-aromatic  
587 polyamide separation layer and its relationship with separation performance of TFC  
588 membranes. *J. Membr. Sci.* **2017**, *541*, 174-188.

589 35. Yan, W.; Wang, Z.; Zhao, S.; Wang, J.; Zhang, P.; Cao, X., Combining co-solvent-  
590 optimized interfacial polymerization and protective coating-controlled chlorination for  
591 highly permeable reverse osmosis membranes with high rejection. *J. Membr. Sci.* **2019**,  
592 *572*, 61-72.

593 36. Hirose, M.; Ito, H.; Kamiyama, Y., Effect of skin layer surface structures on the  
594 flux behaviour of RO membranes. *J. Membr. Sci.* **1996**, *121*, (2), 209-215.

595 37. Tan, Z.; Chen, S.; Peng, X.; Zhang, L.; Gao, C., Polyamide membranes with  
596 nanoscale Turing structures for water purification. *Science* **2018**, *360*, (6388), 518.

597 38. Shen, L.; Hung, W.-s.; Zuo, J.; Zhang, X.; Lai, J.-Y.; Wang, Y., High-performance  
598 thin-film composite polyamide membranes developed with green ultrasound-assisted  
599 interfacial polymerization. *J. Membr. Sci.* **2019**, *570-571*, 112-119.

600 39. Wong, M. C. Y.; Lin, L.; Coronell, O.; Hoek, E. M. V.; Ramon, G. Z., Impact of  
601 liquid-filled voids within the active layer on transport through thin-film composite  
602 membranes. *J. Membr. Sci.* **2016**, *500*, 124-135.

603 40. Fujioka, T.; Oshima, N.; Suzuki, R.; Price, W. E.; Nghiem, L. D., Probing the  
604 internal structure of reverse osmosis membranes by positron annihilation spectroscopy:  
605 Gaining more insight into the transport of water and small solutes. *J. Membr. Sci.* **2015**,  
606 *486*, 106-118.

607 41. Mi, B.; Eaton, C. L.; Kim, J.-H.; Colvin, C. K.; Lozier, J. C.; Mariñas, B. J.,  
608 Removal of biological and non-biological viral surrogates by spiral-wound reverse  
609 osmosis membrane elements with intact and compromised integrity. *Water Research*  
610 **2004**, *38*, (18), 3821-3832.

611 42. Hornstra, L. M.; Rodrigues Da Silva, T.; Blankert, B.; Heijnen, L.; Beerendonk, E.;  
612 Cornelissen, E. R.; Medema, G., Monitoring the integrity of reverse osmosis  
613 membranes using novel indigenous freshwater viruses and bacteriophages.  
614 *Environmental Science: Water Research and Technology* **2019**, *5*, (9), 1535-1544.

615 43. Slipko, K.; Reif, D.; Wögerbauer, M.; Hufnagl, P.; Krampe, J.; Kreuzinger, N.,  
616 Removal of extracellular free DNA and antibiotic resistance genes from water and

617 wastewater by membranes ranging from microfiltration to reverse osmosis. *Water Res.*  
618 **2019**, *164*, 114916.

619 44. Fujioka, T.; Boivin, S., Assessing the passage of particles through polyamide  
620 reverse osmosis membranes. *Sep. Purif. Technol.* **2019**, *226*, 8-12.

621 45. Kwon, Y.-N.; Shih, K.; Tang, C.; Leckie, J. O., Adsorption of perfluorinated  
622 compounds on thin-film composite polyamide membranes. *Journal of Applied Polymer*  
623 *Science* **2012**, *124*, (2), 1042-1049.

624 46. Tang, C. Y.; Kwon, Y.-N.; Leckie, J. O., Probing the nano- and micro-scales of  
625 reverse osmosis membranes—A comprehensive characterization of physiochemical  
626 properties of uncoated and coated membranes by XPS, TEM, ATR-FTIR, and streaming  
627 potential measurements. *J. Membr. Sci.* **2007**, *287*, (1), 146-156.

628 47. Tang, C. Y.; Kwon, Y.-N.; Leckie, J. O., Effect of membrane chemistry and coating  
629 layer on physiochemical properties of thin film composite polyamide RO and NF  
630 membranes: II. Membrane physiochemical properties and their dependence on  
631 polyamide and coating layers. *Desalination* **2009**, *242*, (1), 168-182.

632 48. Do, V. T.; Tang, C. Y.; Reinhard, M.; Leckie, J. O., Degradation of polyamide  
633 nanofiltration and reverse osmosis membranes by hypochlorite. *Environmental Science*  
634 *& Technology* **2012**, *46*, (2), 852-859.

635 49. Cuperus, F. P.; Bargeman, D.; Smolders, K., A new method to determine the skin  
636 thickness of asymmetric UF-membranes using colloidal gold particles. *Journal of*  
637 *Colloid And Interface Science* **1990**, *135*, (2), 486-495.

638 50. Avena, M. J.; Koopal, L. K.; van Riemsdijk, W. H., Proton Binding to Humic Acids:  
639 Electrostatic and Intrinsic Interactions. *J. Colloid Interface Sci.* **1999**, *217*, (1), 37-48.

640 51. Dai, R.; Guo, H.; Tang, C. Y.; Chen, M.; Li, J.; Wang, Z., Hydrophilic selective  
641 nanochannels created by metal organic frameworks in nanofiltration membranes  
642 enhance rejection of hydrophobic endocrine-disrupting compounds. *Environmental*  
643 *Science & Technology* **2019**, *53*, (23), 13776-13783.

644 52. Yip, N. Y.; Tiraferri, A.; Phillip, W. A.; Schiffman, J. D.; Elimelech, M., High  
645 performance thin-film composite forward osmosis membrane. *Environ. Sci. Technol.*  
646 **2010**, *44*, (10), 3812-3818.

647 53. Kurihara, M.; Sasaki, T.; Nakatsuji, K.; Kimura, M.; Henmi, M., Low pressure  
648 SWRO membrane for desalination in the mega-ton water system. *Desalination* **2015**,  
649 *368*, 135-139.

650 54. Shi, M.; Wang, Z.; Zhao, S.; Wang, J.; Zhang, P.; Cao, X., A novel pathway for high  
651 performance RO membrane: Preparing active layer with decreased thickness and  
652 enhanced compactness by incorporating tannic acid into the support. *J. Membr. Sci.*  
653 **2018**, *555*, 157-168.

654 55. Ma, X.-H.; Yang, Z.; Yao, Z.-K.; Guo, H.; Xu, Z.-L.; Tang, C. Y., Interfacial  
655 polymerization with electrosprayed microdroplets: Toward controllable and ultrathin  
656 polyamide membranes. *Environmental Science & Technology Letters* **2018**, *5*, (2), 117-  
657 122.

658 56. Fujioka, T.; O'Rourke, B. E.; Michishio, K.; Kobayashi, Y.; Oshima, N.;  
659 Kodamatani, H.; Shintani, T.; Nghiem, L. D., Transport of small and neutral solutes  
660 through reverse osmosis membranes: Role of skin layer conformation of the polyamide

661 film. *J. Membr. Sci.* **2018**, *554*, 301-308.

662 57. Tang, C. Y.; Kwon, Y.-N.; Leckie, J. O., Fouling of reverse osmosis and  
663 nanofiltration membranes by humic acid—Effects of solution composition and  
664 hydrodynamic conditions. *J. Membr. Sci.* **2007**, *290*, (1), 86-94.

665 58. Coronell, O.; Mariñas, B. J.; Cahill, D. G., Depth heterogeneity of fully aromatic  
666 polyamide active layers in reverse osmosis and nanofiltration membranes. *Environ. Sci.*  
667 *Technol.* **2011**, *45*, (10), 4513-4520.

668 59. Geise, G. M.; Park, H. B.; Sagle, A. C.; Freeman, B. D.; McGrath, J. E., Water  
669 permeability and water/salt selectivity tradeoff in polymers for desalination. *J. Membr.*  
670 *Sci.* **2011**, *369*, (1), 130-138.

671 60. Park, H. B.; Kamcev, J.; Robeson, L. M.; Elimelech, M.; Freeman, B. D.,  
672 Maximizing the right stuff: The trade-off between membrane permeability and  
673 selectivity. *Science* **2017**, *356*, (6343), eaab0530.

674 61. Yao, Z.; Guo, H.; Yang, Z.; Lin, C.; Zhu, B.; Dong, Y.; Tang, C. Y., Reactable  
675 substrate participating interfacial polymerization for thin film composite membranes  
676 with enhanced salt rejection performance. *Desalination* **2018**, *436*, 1-7.

677 62. Li, X.; Li, Q.; Fang, W.; Wang, R.; Krantz, W. B., Effects of the support on the  
678 characteristics and permselectivity of thin film composite membranes. *J. Membr. Sci.*  
679 **2019**, *580*, 12-23.

680 63. Ramon, G. Z.; Wong, M. C. Y.; Hoek, E. M. V., Transport through composite  
681 membrane, part 1: Is there an optimal support membrane? *J. Membr. Sci.* **2012**, *415*-  
682 *416*, 298-305.

683 64. Spiegler, K. S.; Kedem, O., Thermodynamics of hyperfiltration (reverse osmosis):  
684 criteria for efficient membranes. *Desalination* **1966**, *1*, (4), 311-326.

685 65. Urama, R. I.; Mariñas, B. J., Mechanistic interpretation of solute permeation  
686 through a fully aromatic polyamide reverse osmosis membrane. *Journal of Membrane*  
687 *Science* **1997**, *123*, (2), 267-280.

688 66. Coronell, O.; Mi, B.; Mariñas, B. J.; Cahill, D. G., Modeling the Effect of Charge  
689 Density in the Active Layers of Reverse Osmosis and Nanofiltration Membranes on the  
690 Rejection of Arsenic(III) and Potassium Iodide. *Environmental Science & Technology*  
691 **2013**, *47*, (1), 420-428.

692 67. Nghiem, L. D.; Schäfer, A. I.; Elimelech, M., Removal of natural hormones by  
693 nanofiltration membranes: Measurement, modeling, and mechanisms. *Environ. Sci.*  
694 *Technol.* **2004**, *38*, (6), 1888-1896.

695 68. Yang, L.; She, Q.; Wan, M. P.; Wang, R.; Chang, V. W. C.; Tang, C. Y., Removal of  
696 haloacetic acids from swimming pool water by reverse osmosis and nanofiltration.  
697 *Water Research* **2017**, *116*, 116-125.

698 69. Yang, Z.; Zhou, Z.-w.; Guo, H.; Yao, Z.; Ma, X.-h.; Song, X.; Feng, S.-P.; Tang, C.  
699 Y., Tannic Acid/Fe<sup>3+</sup> Nanoscaffold for Interfacial Polymerization: Toward Enhanced  
700 Nanofiltration Performance. *Environmental Science & Technology* **2018**, *52*, (16),  
701 9341-9349.

702 70. Zhou, Z.; Hu, Y.; Boo, C.; Liu, Z.; Li, J.; Deng, L.; An, X., High-performance thin-  
703 film composite membrane with an ultrathin spray-coated carbon nanotube interlayer.  
704 *Environ. Sci. Technol. Lett.* **2018**, *5*, (5), 243-248.



705 71. Yang, X.; Du, Y.; Zhang, X.; He, A.; Xu, Z.-K., Nanofiltration membrane with a  
706 mussel-inspired interlayer for improved permeation performance. *Langmuir* **2017**, *33*,  
707 (9), 2318-2324.  
708