



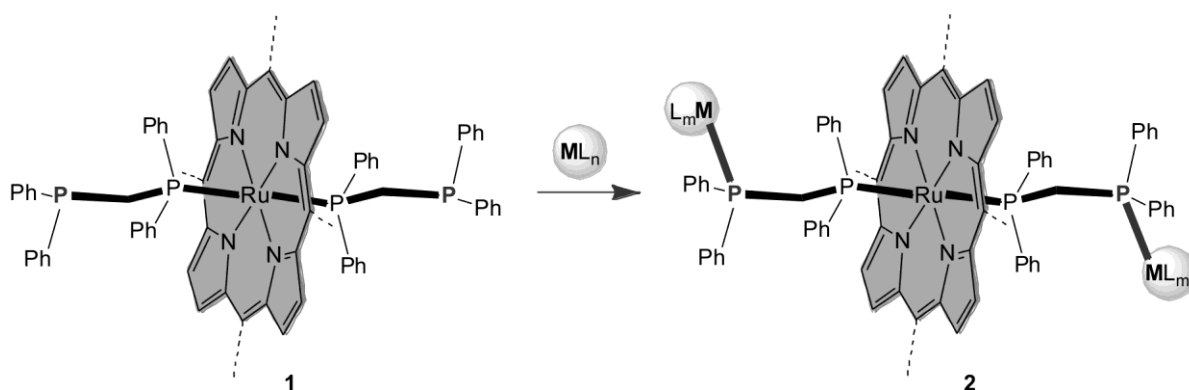
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METALLOPORPHYRIN-INCORPORATED DIPHOSPHINE LIGANDS FOR METAL ION-BINDING

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Diphosphine ligands have been widely used in organometallic chemistry and catalysis.¹ By incorporation of functional units such as metallomacrocycles, the resulting functionalized diphosphines could exhibit unusual properties or binding behavior. In this study, we prepared several examples of ruthenium porphyrin phosphine complexes $[\text{Ru}^{\text{II}}(\text{Por})(\text{dppm})_2]$ (**1**; Por = TTP, 4-MeO-TTP, F₂₀-TTP; dppm = bis(diphenylphosphino)methane) by a similar method to that previously reported for their congeners.² Reaction of complexes **1** with a number of metal complexes ML_n afforded $[(\text{L}_m\text{M})(\mu\text{-dppm})\text{Ru}^{\text{II}}(\text{Por})(\mu\text{-dppm})(\text{ML}_m)]$ (**2**; M = Ag, Au), which have been characterized by spectroscopic methods including ¹H NMR, ³¹P NMR, and UV/Vis spectroscopy, and also by X-ray crystal structure determination. The formation of complexes **2** from complexes **1** demonstrates the role of complexes **1** as a unique type of diphosphine ligands functionalized with metalloporphyrins (which constitute a large family of metal complexes that resemble heme cores in biological systems and exhibit a wide variety of applications³). Studies are underway to explore the properties of this new type of metalloporphyrin-incorporated diphosphine complexes of transition metals.



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